

Name: _____

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Chapter 6 Study Guide: Periodicity

1. Describe the historical development of the periodic table (Newland, Meyer, Mendeleev, Mosely) Describe the contributions made by each of the scientist.

Newland – First to order the elements in a periodic table –developed the law of octaves

Meyers – Ordered elements by atomic mass but is given little credit for the current periodic table

Mendeleev – ordered elements by atomic mass and is given the most credit for the current periodic table because he left holes in his table for undiscovered elements and described their properties

Mosely – ordered elements by atomic number (modern periodic table)

2. Fill in the blanks below:

The elements in the periodic table are arranged in order of increasing Atomic

Number. The vertical columns are called Groups or

Families. The horizontal rows are called Periods. The

elements within the same Family have the same chemical properties due to the number of their Valence Electrons. Metals are located on the

Left side of the periodic table. The nonmetals are located on the

Right side of the periodic table. The metalloids are located on

the Stair step between the metals and nonmetals.

3. Describe the general properties of the following groups of elements on the periodic table:

- a. Alkali Metals

1 valence electrons, metals, highly reactive, s-block

- b. Alkaline Earth metals

2 valence electrons, metals, reactive, s-block

- c. Halogens

7 valence electrons, high reactive, nonmetals, p-block

- d. Noble Gases

8 noble electrons, nonreactive, nonmetals, all gases, p-block

- e. Transition Metals

Variable valence electrons, metals, malleable, ductile, metals, conductive

- f. Inner Transition Metals

Variable valence electrons, metals, f-block

Chapter 6 Periodic Table Answers

JA Banks



Chapter 6 Periodic Table Answers:

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In a world inundated with screens and the cacophony of immediate transmission, the profound energy and mental resonance of verbal artistry often fade in to obscurity, eclipsed by the constant onslaught of sound and distractions. However, nestled within the lyrical pages of **Chapter 6 Periodic Table Answers**, a interesting work of fictional beauty that impulses with organic emotions, lies an remarkable trip waiting to be embarked upon. Written by way of a virtuoso wordsmith, this exciting opus guides readers on a mental odyssey, gently revealing the latent potential and profound affect stuck within the complicated internet of language. Within the heart-wrenching expanse of the evocative evaluation, we can embark upon an introspective exploration of the book is main themes, dissect their captivating writing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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