

Chapter 7 Mid-Chapter Test

(Lectures 7-1 through 7-4)

SCORE _____

Part I Write the letter for the correct answer in the blank at the right of each question.

1. Simplify $(x^2)(x^2)(x^2)(x^2)(x^2)$.

A $x^{10}y^{10}$

B x^5y^5

C x^{10}

D x^{15}

1. _____

2. Simplify $(2a^2b^3)^2(-2a^2b^3)^2$.

F $-12a^4b^6$

G $12a^4b^6$

H $-24a^{10}b^{10}$

J $24a^{10}b^{10}$

2. _____

Simplify. Assume the denominator is not equal to zero.

3. $\frac{m^2n^2}{m^2n^2}$

A $\frac{m^2}{n^2}$

B $-\frac{m^2}{n^2}$

C $-\frac{m^2}{n^2}$

D $\frac{m^2}{n^2}$

3. _____

4. $\frac{(2x^2y^3)^2}{(2x^2y^3)^2}$

F $\frac{2}{3}$

G $\frac{m^2}{n^2}$

H x

J $\frac{2}{3}$

4. _____

5. Solve $2^{3x+10} = 128$.

A 3

B 2

C 1

D -1

5. _____

6. Express 0.000002 in scientific notation.

F 7.02×10^{-7}

H 7.02×10^{-6}

G 7.02^{-7}

J 7.02×10^{-6}

6. _____

7. Evaluate $\frac{4.08 \times 10^7}{3.9 \times 10^{12}}$.

A 1.6×10^5

B 1.6×10^6

C 2.3104×10^{19}

D 2.3104×10^{18}

7. _____

Part II Evaluate each product. Express the results in both scientific notation and standard form.

8. $(4.2 \times 10^5)(1.6 \times 10^{-3})$

8. _____

9. $(7.8 \times 10^{-2})(4 \times 10^{-4})$

9. _____

Simplify. Assume that no denominator is equal to zero.

10. $\frac{2x^2y^2}{2x^2y^2}$

11. $\frac{-12a^{-1}b^2c}{-12a^{-1}b^2c}$

10. _____

Solve each equation.

12. $6^x = 16$

12. _____

13. $216^{x+5} = 6$

13. _____

Chapter 7 Answers Continued

Albert A Gayle



Chapter 7 Answers Continued:

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively c 6 h 8 o 6 2h⁺ 2no²⁻ c 6 h 6 o 6 2h⁺ 2 o²⁻ no choose 1 answer c 6 h 8 o 6 and no 2 a c 6 h 8 o 6 and no 2 no 2 and c 6 h 8 o 6 b no 2 and c 6 h 8 o 6 h and no 2 c h and

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