

Physical Science Chapter 66 (Properties of Atoms and the Periodic Table) Worksheet 1

Name: \_\_\_\_\_

Period: \_\_\_\_\_

Use your notebook to define the following terms

Atom: \_\_\_\_\_

Atomic number: \_\_\_\_\_

Average atomic mass: \_\_\_\_\_

Atomic mass: \_\_\_\_\_

Electron: \_\_\_\_\_

Electron cloud: \_\_\_\_\_

Electron dot diagram: \_\_\_\_\_

Group: \_\_\_\_\_

Isotope: \_\_\_\_\_

Mass number: \_\_\_\_\_

Neutron: \_\_\_\_\_

Nucleus: \_\_\_\_\_

Period: \_\_\_\_\_

Periodic table: \_\_\_\_\_

Proton: \_\_\_\_\_

Quark: \_\_\_\_\_

Multiple choice (Choose the best answer)

- 1) In which state of matter are most of the elements to the left of the stair-step line in the periodic table?
  - a) Gas
  - b) Liquid
  - c) Plasma
  - d) Solid
  
- 2) Which is a name for a pattern that repeats?
  - a) Isotope
  - b) Mendeleev
  - c) Periodic
  - d) Transition
  
- 3) Which of the following methods would be most useful for describing atoms?
  - a) Visualizing with microscopes
  - b) Observing in a test tube
  - c) Constructing a model
  - d) Touching with bare hands
  
- 4) An atom containing six positive charges and six negative charges has \_\_\_\_\_ charge.
  - a) a positive
  - b) a negative
  - c) a strong negative
  - d) no net

# Chapter Test Properties Of Atoms The Periodic Table Answer Key

**Stuart J. Baum, C. W. J. Scaife**

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