

Chapter 8 Solutions, Acids, and Bases

Section 8.2 Solubility and Concentration

(pages 235–239)

*This section explains solubility, the factors affecting solubility, and different ways of expressing the concentration of a solution.***Reading Strategy (page 235)**

Previewing Before you read the section, rewrite the topic headings as *how*, *why*, and *what* questions. As you read, write an answer to each question. For more information on this reading strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Question	Answer
What is solubility?	
	Solvent, temperature, and pressure

Solubility (pages 235–237)

- Define solubility. _____

- List the following solutes in order from most soluble to least soluble in water: table salt, baking soda, table sugar.
 - _____
 - _____
 - _____
- Circle the letters that identify how solutions can be classified based on solubility.

a. unsaturated	b. desaturated
c. saturated	d. supersaturated
- Describe a saturated solution. _____

- A solution that has less than the maximum amount of solute that can be dissolved is called a(n) _____.
- Is the following sentence true or false? It is impossible for a solution to contain more solute than the solvent can hold at a given temperature. _____

Chapter 8 Solutions Section 3 Solubility And Concentration

M Woodhall



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updated to reflect the enormous advances in molecular and protein structure Features a new chapter on nucleic acids gene expression and recombinant DNA technology as well as a new chapter on nucleotide metabolism Integrated Biochemical Interactions CD Chemistry Oxtoby,1994 *Recovery from Dilute Solutions* Michael Charles Milner Cockrem,1987

Introduction to Geochemistry Konrad Bates Krauskopf,Dennis K. Bird,1995 Throughout the book attention is continually directed to the relations between theoretical formulas and results of controlled laboratory experiments as well as to geologic field observations The book begins with an introduction to chemical equilibrium concentrating on the carbonate and silicate equilibria that are important in geologic environments Next comes a brief look at the chemistry of crystalline solids and reactions at mineral surfaces The Physical Chemistry of Electrolytic Solutions Herbert Spencer Harned,Benton Brooks Owen (1900- joint author),1943 **Sci-tech Book Profiles** ,1965 Includes title page table of contents list of contributors preface and all indexes of each book Dissolution and Condensation Kinetics of Crystalline and Amorphous Silica in Alkaline Solutions Shawn David Thornton,1985

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