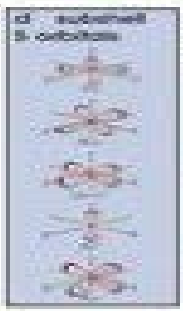


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ATOMIC STRUCTURE

Atomic Number, Z , of an element is the number of protons in the nucleus. For a neutral atom, $Z = +$ the number of electrons. The mass number, A , is the sum of protons and neutrons in the nucleus. Isotopes of an element have the same Z , different A . The atomic mass is the weighted average of isotopes. The atomic weight is the weighted average of isotopes for a given element.

ATOMIC QUANTUM NUMBERS AND ORBITALS

Principal quantum number, n , is the number of shells. $n = 1, 2, 3, \dots$. Subshell quantum numbers are $l = 0, 1, 2, \dots, n-1$. Magnetic quantum number, m_l , is the number of orbitals in a subshell. Spin quantum number, m_s , is the number of electrons in an orbital.

NUCLEAR CHEMISTRY

Nuclear reactions occur when particles (alpha, beta, gamma, neutron, proton) are emitted from the nucleus. Radioactive decay is the process by which an unstable nucleus loses energy by emitting radiation. The half-life of a radioactive substance is the time it takes for half of the substance to decay.

MANY-ELECTRON ATOMS

Orbitals and quantum numbers are used to describe the arrangement of electrons in an atom. The Aufbau principle states that electrons fill orbitals from lowest to highest energy. Hund's rule states that electrons fill degenerate orbitals singly before pairing up. The Pauli exclusion principle states that no two electrons can have the same set of quantum numbers.

n	l	m _l	m _s	Orbital
1	0	0	±1/2	1s
2	0	0	±1/2	2s
2	1	0	±1/2	2p _z
2	1	±1	±1/2	2p _x , 2p _y
3	0	0	±1/2	3s
3	1	0	±1/2	3p _z
3	1	±1	±1/2	3p _x , 3p _y
3	2	0	±1/2	3d _{z²}
3	2	±1	±1/2	3d _{xy} , 3d _{xz} , 3d _{yz}

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Mark Jackson



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page laminated guide is intended to serve as a reminder and quick study review for all aspects of the chemistry lab from safety to chemicals to instruments and safety again first last and always Designed to find answers fast and to be spill proof it is perfect for high school and early college students studying chemistry as well as anyone interested in learning how scientists safely study the molecules that make up everything around us 6 page laminated guide includes Lab Safety Training General Lab Guidelines Working With Chemicals Exposure to Chemicals First Aid Know Your Lab Reagents Chemical Spills Waste Management Useful Chemical Information Periodic Table Included Lab Must Knows Safe Use of Lab Equipment Data Manipulation Preparing a Solution Inorganic Chemistry Mark Jackson,2010-05-31 Our 3 panel 6 page guide is jam packed with chemistry information not covered in organic chemistry and biochemistry Perfect for science students who want to focus on the aspects of chemistry that go beyond what is covered in material that deals with organic topics this guide covers the key concepts principles figures and formulas that inorganic chemistry students will need to know in order to succeed Tables images and graphic elements further enhance the text **Chemistry 2** BarCharts, Inc,Jane P. Gardner,2016 QuickStudy Academic Mark Jackson (PhD and Science Professor),2015 Quick Reference for the core essentials of a subject and class that is challenging at best and that many students struggle with In 6 laminated pages our experienced chemistry author and professor gathered key elements organized and designed to use along with your text and lectures as a review before testing or as a memory companion that keeps key answers always at your fingertips **AP Chemistry** BarCharts, Inc.,2010-12-31 Preparation for the AP Chemistry exam has never been easier once BarCharts gets involved All the latest information regarding this important exam is featured in our jam packed 3 panel guide including sample problems that tackle each area of the test Tables graphs and full color charts and diagrams are also featured to help give students the winning edge they need *Periodic Table Basic* Mark Jackson,2018 **Physics Terminology** BarCharts, Inc,Mark Jackson,2013-12-31 Essential terminology for Physics study pulled together in one guide to maximize success in College and High School courses Succinct definitions by our resident Harvard Ph D Chemistry author and professor ensure the usefulness of this handy guide from high school to college *Biology* ,1999 **Algebra** S. B. Kizlik,2006-11-30 Essence of the course problems answers formulas equations full color graphics Cover **Periodic Table Basic** Mark Jackson,2002-07-16 Periodic table of the elements **The Public-school Journal** ,1892 **Physical Science - Physics and Chemistry** Jane Gardner,2022-11 Essential quick reference for a physical science class covering physics and chemistry which would generally be high school level or if combined with our Physical Science 2 guide together would cover a college level Physical Science course Expertly and succinctly written by author STEM curriculum developer and professor Jane Parks Gardner MSc MScEd in an easily accessible outline format with color coded sections This 6 page laminated guide is overflowing with the facts you need to know to boost confidence test scores and grades The value at this price is unmatched for a study tool that can be a review and refresher through both high school and college 6 page laminated guide includes Scientific Method Measurement

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 - The Rise of Digital Reading Chemistry Quickstudy Reference Guides Academic
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Quickstudy Reference Guides Academic
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Quickstudy Reference Guides Academic
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Quickstudy Reference Guides Academic
 - Personalized Recommendations
 - Chemistry Quickstudy Reference Guides Academic User Reviews and Ratings
 - Chemistry Quickstudy Reference Guides Academic and Bestseller Lists
5. Accessing Chemistry Quickstudy Reference Guides Academic Free and Paid eBooks
 - Chemistry Quickstudy Reference Guides Academic Public Domain eBooks
 - Chemistry Quickstudy Reference Guides Academic eBook Subscription Services
 - Chemistry Quickstudy Reference Guides Academic Budget-Friendly Options
6. Navigating Chemistry Quickstudy Reference Guides Academic eBook Formats

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 - Chemistry Quickstudy Reference Guides Academic Compatibility with Devices
 - Chemistry Quickstudy Reference Guides Academic Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Quickstudy Reference Guides Academic
 - Highlighting and Note-Taking Chemistry Quickstudy Reference Guides Academic
 - Interactive Elements Chemistry Quickstudy Reference Guides Academic
 8. Staying Engaged with Chemistry Quickstudy Reference Guides Academic
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Quickstudy Reference Guides Academic
 9. Balancing eBooks and Physical Books Chemistry Quickstudy Reference Guides Academic
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Quickstudy Reference Guides Academic
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemistry Quickstudy Reference Guides Academic
 - Setting Reading Goals Chemistry Quickstudy Reference Guides Academic
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemistry Quickstudy Reference Guides Academic
 - Fact-Checking eBook Content of Chemistry Quickstudy Reference Guides Academic
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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 - Integration of Multimedia Elements
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