

Answer \_\_\_\_\_

**MULTIPLE CHOICE** Choose the one alternative that best completes the statement or answers the question.

1. Which of the following compounds will give a color reaction when added to an alkaloid? 10 \_\_\_\_\_

- (A) Resorcinol (1,3-dihydroxybenzene)
- (B) 2,6-Dichlororesorcinol (2,6-dichloro-1,3-dihydroxybenzene)
- (C) Resorcinol (1,3-dihydroxybenzene)
- (D) Vaniline (3-hydroxybenzaldehyde)
- (E) None of these compounds should be suitable to alkaloid.

2. Which of the following is the correct (skeletal) structure of Cocaine? 10 \_\_\_\_\_

- (A) 
- (B) 
- (C) 
- (D) 
- (E) None of these are correct representations of Cocaine.

3. Which of the following ions should have the most significant hydration enthalpy? 10 \_\_\_\_\_

- (A)  $\text{Na}^+$
- (B)  $\text{Ca}^{2+}$
- (C)  $\text{Mg}^{2+}$
- (D)  $\text{Zn}^{2+}$
- (E)  $\text{Ag}^+$

4. Determine the  $\text{pH}$  of a 0.10 M solution of the  $\text{H}_2\text{A}$  is a  $\text{K}_1 = 1.0 \times 10^{-4}$  and the  $\text{K}_2 = 1.0 \times 10^{-8}$ . 10 \_\_\_\_\_

- (A) 0.50
- (B) 1.00
- (C) 4.00
- (D) 7.00
- (E) 13.00

5. Which of the following elements is generally "basic"? 10 \_\_\_\_\_

- (A) Strontium (element in highly electropositive period 5 transition metal)
- (B) Strontium (element in most electropositive metal region of periodic table)
- (C) Strontium (element in highly electronegative period)
- (D) Strontium (element in highly electronegative period)
- (E) Strontium (element)

6. Which of the following elements is "acidic"? 10 \_\_\_\_\_

- (A) Strontium (element in highly electronegative period 5 transition metal)
- (B) Strontium (element in most electronegative metal region of periodic table)
- (C) Strontium (element in most electronegative metal region of periodic table)
- (D) Strontium (element in most electronegative metal region of periodic table)
- (E) Strontium (element in most electronegative metal region of periodic table)

7. Which of the following should have the largest atomic ion hydration enthalpy? 10 \_\_\_\_\_

- (A)  $\text{Na}^+$
- (B)  $\text{Ca}^{2+}$
- (C)  $\text{Mg}^{2+}$
- (D)  $\text{Zn}^{2+}$
- (E)  $\text{Ag}^+$

# Chemistry Chapter 12 Mixed Review Answers

**Neil D. Jespersen, Alison Hyslop**



## **Chemistry Chapter 12 Mixed Review Answers:**

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