

Name _____ Date _____ Class _____

CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Matter—Properties and Changes

Section 3.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substance
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Matter is anything with (1) mass and volume. A(2) substance is a form of matter with a uniform and unchanging composition.Substances have specific, unchanging (3) properties that can be observed.Substances have both physical and chemical properties. (4) Physical

properties can be observed without changing a substance's chemical composition. Color,

hardness, and (5) density are examples. Other properties cannot be

observed without changing the composition of a substance. These are called

(6) chemical properties. An example is the tendency of iron to form

rust when exposed to air.

Label each property as either physical or chemical.

- physical 7. Chemical formula H_2O
- chemical 8. Irons gain cohesion when exposed to heat or
- chemical 9. Remains unchanged when in the presence of nitrogen
- physical 10. Crystals
- physical 11. Solid at normal temperatures and pressures
- chemical 12. Ability to combine with another substance
- physical 13. Melting point
- physical 14. Liquid at normal temperatures and pressures
- physical 15. Boiling point is $100^\circ C$
- physical 16. Conducts electricity
- physical 17. Density is $\frac{1}{8} \frac{g}{cm^3}$

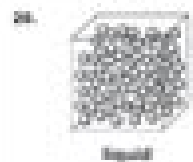
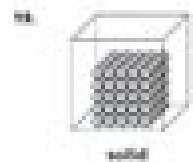
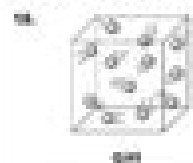
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CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Section 3.1 Matter

In your textbook, read about states of matter.

Label each drawing with one of these words: solid, liquid, gas.



For each statement below, write true or false.

- true 21. All matter that we encounter in everyday life exists in one of three physical states.
- true 22. A solid has definite shape and volume.
- false 23. A liquid has a definite shape and takes on the volume of its container.
- true 24. A gas has both the shape and the volume of its container.
- false 25. The particles in a gas cannot be compressed into a smaller volume.
- false 26. Liquids tend to contract when heated.
- false 27. The particles in a solid are spaced far apart.
- false 28. The words gas and vapor can be used interchangeably.

Chemistry Matter And Change Chapter 6 Assessment Answers

Joacim Rocklöv



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Chemistry John S. Phillips,Cheryl Wistrom,2000

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Chapter Resource 2 Chemistry of Life

Biology Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004 **Biology** Kenneth Raymond Miller,Prentice Hall (School Division),1999-02

Canadian Acid Deposition Science Assessment, 2004 Meteorological Service of Canada,2005 The 2004 Canadian Acid Deposition Science Assessment has been prepared by researchers from the federal and provincial governments and academia This assessment builds upon the results presented in the 1997 Canadian Acid Rain Assessment The assessment consists of two documents the Summary of Key Results and the Assessment The Summary of Key Results synthesizes the major findings of the Assessment in the context of ten questions and subquestions put forth by the acid deposition policy and science communities The Assessment presents a comprehensive review of acid deposition science in Canada Environment Canada Website *Michigan School Moderator* ,1893 Government Reports Announcements & Index ,1996-07 **Government Reports Annual Index** ,1994 Holt Chemistry Ralph Thomas Myers,2004 Chemical Building Blocks Michael J. Padilla,David V. Frank,John G. Little,Steve Miller,2000 Science ,2001 **Merrill Earth Science** Ralph M. Feather,Susan Leach Snyder,Dale T. Hesser,1995 *Biology: How Life Works (Volume 1)* James Morris,2013-01-04 Ordinarily textbooks are developed by first writing chapters then making decisions about art and images and finally once the book is complete assembling a test bank and ancillary media This process dramatically limits the integration across resources and reduces art media and assessments to ancillary material rather than essential resources for student learning Biology How Life Works is the first project to develop three pillars the text the visual program and the assessment at the same time All three pillars were developed in parallel to make sure that each idea is addressed in the most appropriate medium and to ensure authentic integration These three pillars are all tied to the same set of core concepts share a common language and use the same visual palette In this way the text visual program and assessments are integral parts of student learning rather than just accessories to the text

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