

Reference Tables for Physical Setting/CHEMISTRY

2002 Edition

Table A
Standard Temperature and Pressure

| Name | Value | Unit |
|----------------------|--------------------|--------------------------|
| Standard Pressure | 101.3 kPa 1 atm | kilopascal atmosphere |
| Standard Temperature | 273 K 0°C | kelvin degree Celsius |

Table B
Physical Constants for Water

| | |
|------------------------------------------------|-------------|
| Heat of Fusion | 334 J/g |
| Heat of Vaporization | 2260 J/g |
| Specific Heat Capacity of H ₂ O (l) | 4.18 J/g•°C |

Table C
Selected Prefixes

| Factor | Prefix | Symbol |
|-------------------|--------|--------|
| 10 ³ | kilo- | k |
| 10 ⁻¹ | deci- | d |
| 10 ⁻² | centi- | c |
| 10 ⁻³ | milli- | m |
| 10 ⁻⁶ | micro- | μ |
| 10 ⁻⁹ | nano- | n |
| 10 ⁻¹² | pico- | p |

Table D
Selected Units

| Symbol | Name | Quantity |
|--------|------------------|--------------------------------|
| m | meter | length |
| g | gram | mass |
| Pa | pascal | pressure |
| K | kelvin | temperature |
| mol | mole | amount of substance |
| J | joule | energy, work, quantity of heat |
| s | second | time |
| L | liter | volume |
| ppm | part per million | concentration |
| M | molarity | solution concentration |

Table E
Selected Polyatomic Ions

| | | | |
|-----------------------------------------------------------------------------------------------|--------------------|----------------------------------------------|------------------|
| H ₃ O ⁺ | hydronium | CrO ₄ ²⁻ | chromate |
| Hg ₂ ²⁺ | mercurous (II) | Cr ₂ O ₇ ²⁻ | dichromate |
| NH ₄ ⁺ | ammonium | MnO ₄ ⁻ | permanganate |
| C ₂ H ₃ O ₂ ⁻ CH ₃ COO ⁻ | acetate | NO ₂ ⁻ | nitrite |
| | | NO ₃ ⁻ | nitrate |
| CN ⁻ | cyanide | O ₂ ²⁻ | peroxide |
| CO ₃ ²⁻ | carbonate | OH ⁻ | hydroxide |
| HCO ₃ ⁻ | hydrogen carbonate | PO ₄ ³⁻ | phosphate |
| C ₂ O ₄ ²⁻ | oxalate | SCN ⁻ | thiocyanate |
| ClO ⁻ | hypochlorite | SO ₃ ²⁻ | sulfite |
| ClO ₂ ⁻ | chlorite | SO ₄ ²⁻ | sulfate |
| ClO ₃ ⁻ | chlorate | H ₂ SO ₄ | hydrogen sulfate |
| ClO ₄ ⁻ | perchlorate | S ₂ O ₃ ²⁻ | thiosulfate |

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