

STOICHIOMETRY WORKSHEET

1. Calculate the number of moles of NaOH that are needed to react with 500 g of H₂SO₄ according to the following equation:



2. Calculate the mass of NH₃ that can be produced from the reaction of 125 g of NCl₃ according to the following equation:



3. Identify the limiting reactant and determine the mass of CO₂ that can be produced from the reaction of 25 g of C₃H₈ with 75 g of O₂ according to the following equation:



4. How many grams of SO₂ are produced when 152 g of CS₂ reacts with 48.0 g of O₂ according to the following equation:



Chemistry Semester 1 Unit 9 Stoichiometry Answers

Stanford University



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Instructor's Manual Michell J. Sienko, Robert A. Plane, 1966 *Stanford University Bulletin* Stanford University, 1991
Mapping College Chemistry Stephen DeMeo, 2019-03-01 This text is a chemistry problem solving resource appropriate for teachers and their students who are enrolled in high school Advanced Placement Chemistry or in a first year college General Chemistry course The book incorporates a chemistry problem solving plan one that uses an innovative graphic organizer strategy The strategy successfully evaluated with students combines problem solving processes with chemical concepts that will allow students to solve the most common and difficult problems encountered in the first year of chemistry Topical problem solving will focus on limiting reactant stoichiometry identifying types of chemical reactions equilibrium acid base equilibria and electrochemistry Why would this resource be of interest to chemistry students To be successful to get into a well known college medical school physical therapy or graduate program often requires that students get an A in your pre requisite Introductory General Chemistry course To make matters worse many college professors feel that only a few students should get A grades and therefore they give difficult exams that many students fail this is the weeding out process that every pre health student is apprehensive about To succeed in this competitive environment entails not just studying harder or longer it means re organizing textbook content so that it is meaningful to the student This is the first text of its kind to employ a reliable research based strategy that incorporates a decision based visual tool to solve chemistry textbook problems ones that can make or break a career **The Software Encyclopedia 2001** ,2001 Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes **The Science Teacher** ,2008 General Catalog - Texas Tech University Texas Tech University, 1944 College Credit Recommendations ,1996 **New Technical Books** New York Public Library, 1970
Agents of Change Clarence Walter Mangham, 1995 Volume one is the first of a series of historical documents designed to provide information on instructional improvement projects that have received grants and loans from the California Community College Fund for Instructional Improvement Chemical Engineering Education ,2004 **Choice** ,1974

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