

Section 12.1

OBJECTIVE 1

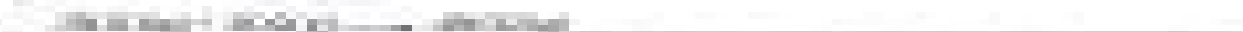
Write complete balanced redox reactions that are indicated.

1. _____
 - a. _____
 - (a) $\text{Fe}^{2+} + \text{H}_2\text{O}_2 + \text{H}^+ \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}$
 - (b) $\text{Mn}^{2+} + \text{H}_2\text{O}_2 + \text{OH}^- \rightarrow \text{MnO}_2 + \text{H}_2\text{O}$
 - (c) $\text{Cr}^{3+} + \text{H}_2\text{O}_2 + \text{OH}^- \rightarrow \text{CrO}_4^{2-} + \text{H}_2\text{O}$
 - (d) $\text{Fe}^{2+} + \text{H}_2\text{O}_2 + \text{OH}^- \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}$
 - b. _____
 - (a) $\text{Fe}^{2+} + \text{H}_2\text{O}_2 + \text{H}^+ \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}$
 - (b) $\text{Mn}^{2+} + \text{H}_2\text{O}_2 + \text{OH}^- \rightarrow \text{MnO}_2 + \text{H}_2\text{O}$
 - (c) $\text{Cr}^{3+} + \text{H}_2\text{O}_2 + \text{OH}^- \rightarrow \text{CrO}_4^{2-} + \text{H}_2\text{O}$
 - (d) $\text{Fe}^{2+} + \text{H}_2\text{O}_2 + \text{OH}^- \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}$
2. _____
 - a. _____
 - (a) $\text{Fe}^{2+} + \text{H}_2\text{O}_2 + \text{H}^+ \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}$
 - (b) $\text{Mn}^{2+} + \text{H}_2\text{O}_2 + \text{OH}^- \rightarrow \text{MnO}_2 + \text{H}_2\text{O}$
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3. _____
 - a. _____
 - (a) $\text{Fe}^{2+} + \text{H}_2\text{O}_2 + \text{H}^+ \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}$
 - (b) $\text{Mn}^{2+} + \text{H}_2\text{O}_2 + \text{OH}^- \rightarrow \text{MnO}_2 + \text{H}_2\text{O}$
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 - (d) $\text{Fe}^{2+} + \text{H}_2\text{O}_2 + \text{OH}^- \rightarrow \text{Fe}^{3+} + \text{H}_2\text{O}$

4. Write the net ionic equation for the following balanced equation:



a. Balance the equation:



b. Write the net ionic equation for $\text{Fe}^{2+} + \text{HNO}_3$.

c. In a solution of Fe^{2+} ions, how many moles of H^+ must be added to precipitate?

d. What is the molar mass of Fe^{2+} ions in a solution of Fe^{2+} ions?

e. What is the molar mass of Fe^{2+} ions in a solution of Fe^{2+} ions?

Chapter 9 Section 1 Stoichiometry Answers

M Mosston



Chapter 9 Section 1 Stoichiometry Answers:

Reviewing **Chapter 9 Section 1 Stoichiometry Answers**: Unlocking the Spellbinding Force of Linguistics

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