

# CHEM 121: Final Exam Study Guide

## Chapter 1

- Know the Scientific method
- Know the definitions for hypothesis, scientific law, and scientific theory.

## Chapter 2

- Length, mass, weight, volume
  - Know  $1 \text{ cm}^3 = 1 \text{ mL}$  and  $1 \text{ dm}^3 = 1 \text{ L}$
- Significant figures and Scientific notation
  - Rounding, in addition or subtraction, in multiplication or division
  - Keep as many sig figs until final answer
- Scientific notation
- Metric system
  - Know the metric prefixes: kilo (k), deci (d), centi (c), milli (m), and micro ( $\mu$ )
  - Be able to perform metric-metric conversions using these prefixes.
- Use the metric-English conversions provided ( $1 \text{ in.} = 2.54 \text{ cm}$ ;  $1 \text{ lb} = 453.6 \text{ g}$ ;  $1 \text{ qt} = 946 \text{ mL}$ )
- Volume by calculation
  - $V_{\text{rectangular solid}} = \text{length} \times \text{width} \times \text{thickness}$
- Volume by displacement
- Density:  $d = \frac{m}{V}$ 
  - Calculate density, mass, or volume
  - Identify what items sink or float given densities of liquids and solids.
- Temperature
  - Know the formulas for converting "F-to-"C or "C-to-"F and K-to-"C or "C-to-K
- Percentage: ratio of parts per 100 parts
  - Given amount of part and whole, calculate %
  - Use a given % to solve for part or whole

## Chapter 3

- Know that matter is studied at the macroscopic, microscopic, particulate (molecular) levels
- Physical states of matter
  - Determine physical state of substances (solids, liquids, gases) given descriptions of volume, shape, particles moving, etc.
- Classify properties and changes as physical or chemical
  - Know terms for changes of state:
    - Melting, freezing, vaporizing, condensation, sublimation, deposition
- Classification of matter
  - Given examples, determine which are elements, compounds, or mixtures
  - Given molecular-level images, determine which are elements, compounds, or mixtures and solids, liquids, or gases
  - Distinguish between homogeneous and heterogeneous mixtures
- Chemical reaction:
  - reactants: starting materials
  - products: substances produced in reaction

# Chemistry 121 Final Exam Study Guide

**JG Myers**



## **Chemistry 121 Final Exam Study Guide:**

## Reviewing **Chemistry 121 Final Exam Study Guide**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Chemistry 121 Final Exam Study Guide**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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