

## Distributive Property Word Problems

The **distributive property** is a tool to make multiplication with larger numbers easier.

To use the distributive property:  
Break one factor into two addends,  
multiply both addends by the other  
factor, and add together both products.

Break up  
the bigger  
number  
into two  
addends

$$\begin{aligned} & 16 \times 5 \\ & (10 + 6) \times 5 \\ & (10 \times 5) + (6 \times 5) \\ & 50 + 30 = 80 \end{aligned}$$

$$16 \times 5 = 80$$



**Directions:** Solve each problem below using the distributive property. Show your work.

- |  |  |
|--|--|
| <p>1. Shanna made popsicles on a hot summer day. She made 8 trays of popsicles. Each tray makes 16 popsicles. How many popsicles did she make?</p>       | <p>2. Saijeen sold 12 cups of lemonade each hour at her lemonade stand. She was outside for a total of 4 hours. How many cups of lemonade did she sell?</p>                    |
| <p>3. There are 14 crackers in each package. If the family brought 6 packages on the road trip, how many crackers did they have?</p>                     | <p>4. Jeremy gets french fries everyday for lunch at school. If he eats 26 french fries each day, how many fries does he eat in a school week?</p>                             |
| <p>5. The soccer league is holding 5 camps this summer. There will be 15 players in each camp. How many players will attend soccer camp this summer?</p> | <p>6. The chef needed eggs for the new meal on the menu. Each egg carton has 18 eggs. She ordered 9 cartons of eggs. How many eggs did the chef need for her special meal?</p> |



# Distributive Property Word Problem

**CO Houle**

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular glow behind it.

**Distributive Property Word Problem:**

Ignite the flame of optimism with is motivational masterpiece, Fuel Your Spirit with **Distributive Property Word Problem** . In a downloadable PDF format ( Download in PDF: \*), this ebook is a beacon of encouragement. Download now and let the words propel you towards a brighter, more motivated tomorrow.

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web the molar volume will be the volume occupied by one molar mass 32 g of the liquid expressing the density in liters instead of ml we have  $v = \frac{m}{d} = \frac{32 \text{ g}}{1.790 \text{ g/l}} = 17.88 \text{ l}$

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web apr 22 2020 molar volume of gases 1 counting in moles one mole is defined as 6.02 x 10<sup>23</sup> particles this is the avogadro number an unchanging constant whether you are counting hydrogen atoms carbon dioxide molecules or carbonate ions a mole is always 6.02 x 10<sup>23</sup> but as different particles have different mass their molar mass varies

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web the molar volume has the si unit of cubic metres per mole m<sup>3</sup> mol although it is more typical to use the units cubic decimetres per mole dm<sup>3</sup> mol for gases and cubic centimetres per mole cm<sup>3</sup> mol for liquids and solids

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