

## INORGANIC CHEMISTRY

451. Nitrogen can be obtained from air by removing  
 (a) Oxygen (b) Hydrogen (c) Carbon dioxide (d) Both (a) and (c)
452. Which of the following carbide is a methanide?  
 (a)  $\text{Be}_2\text{C}$  (b)  $\text{CaC}_2$  (c)  $\text{Al}_2\text{C}_3$  (d)  $\text{Mg}_2\text{C}_2$
453. Caro's acid is  
 (a)  $\text{H}_2\text{SO}_3$  (b)  $\text{H}_2\text{S}_2\text{O}_8$  (c)  $\text{H}_2\text{SO}_4$  (d)  $\text{H}_2\text{S}_2\text{O}_7$
454. Inorganic benzene is  
 (a)  $\text{B}_3\text{H}_3\text{N}_3$  (b)  $\text{BH}_2\text{NH}_2$  (c)  $\text{B}_3\text{H}_6\text{N}_3$  (d)  $\text{H}_3\text{B}_3\text{N}_3$
455. For the reaction,  $\text{P}(\text{White}) \longrightarrow \text{P}(\text{Red})$ ;  $\Delta H = -18.4 \text{ kJ}$ . It follows that  
 (a) Red P is readily formed from white P  
 (b) White P is readily formed from red P  
 (c) White P can not be converted to red P  
 (d) White P can be converted into red P and red P is more stable
456. Nitrogen dioxide cannot be obtained by heating  
 (a)  $\text{KNO}_3$  (b)  $\text{Pb}(\text{NO}_3)_2$  (c)  $\text{Cu}(\text{NO}_3)_2$  (d)  $\text{AgNO}_3$
457. Reactivity of borazole is greater than that of benzene because:  
 (a) Borazole is a non-polar compound  
 (b) Borazole is a polar compound  
 (c) Borazole is an electron deficient compound  
 (d) Of localized electrons in it
458. When orthophosphoric acid is heated to  $600^\circ \text{C}$ , the product formed is  
 (a) Phosphine,  $\text{PH}_3$  (b) Phosphorus pentoxide,  $\text{P}_2\text{O}_5$   
 (c) Phosphorus acid,  $\text{H}_3\text{PO}_3$  (d) Metaphosphoric acid,  $\text{HPO}_3$
459.  $\text{P}_4 + 3\text{NaOH} + 3\text{H}_2\text{O} \rightarrow \text{A} + 3\text{NaH}_2\text{PO}_2$ , here, 'A' is  
 (a)  $\text{NH}_3$  (b)  $\text{PH}_3$  (c)  $\text{H}_2\text{PO}_2$  (d)  $\text{H}_3\text{PO}_4$
460.  $\text{SO}_2$  in the solid phase exists as  
 (a) linear chain or cyclic trimer (b) planar triangular unit  
 (c) square planar unit (d) none of these
461. Sulphur does not exist as  $\text{S}_2$  molecule because  
 (a) It is less electronegative (b) Does not form  $\text{pi-pi}$  bond  
 (c) It has ability to exhibiting catenation (d) Has tendency to show variable oxidation state
462. Which of the following has S-S BOND?  
 (a)  $\text{H}_2\text{S}_2\text{O}_8$  (b)  $\text{H}_2\text{S}_2\text{O}_7$  (c)  $\text{H}_2\text{S}_2\text{O}_6$  (d) Mustard gas
463. Which of the following is correct order of stability order?  
 (a)  $\text{Pb}^{2+} > \text{Pb}^{4+}$  (b)  $\text{Pb}^{4+} > \text{Pb}^{2+}$  (c)  $\text{Si}^{2+} > \text{Si}^{4+}$  (d)  $\text{Sn}^{4+} > \text{Sn}^{2+}$
464. Quartz is a crystalline variety of  
 (a) Si (b)  $\text{SiO}_2$  (c)  $\text{Na}_2\text{SiO}_3$  (d) SiC
465. The equivalent weight of phosphoric acid in the reaction  
 $\text{NaOH} + \text{H}_3\text{PO}_4 \longrightarrow \text{NaH}_2\text{PO}_4 + \text{H}_2\text{O}$   
 (a) 25 (b) 49 (c) 59 (d) 98

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