

Dehydration Synthesis: Paper Modeling Activity

For each reaction you will be provided with the substance formula and the name of each reactant and product. Remember that dehydration synthesis is a reaction in which two molecules combine to form a larger molecule. Water is a byproduct of this process. Remember that dehydration synthesis is a reaction in which two molecules combine to form a larger molecule. Water is a byproduct of this process. Remember that dehydration synthesis is a reaction in which two molecules combine to form a larger molecule. Water is a byproduct of this process.

Part A: Carbohydrates

1. What are monosaccharides and disaccharides?
What is the molecular formula for each monosaccharide?



What is the molecular formula for this disaccharide?



2. Draw the structure of the disaccharide. Put the C-OH on the carbon of the molecule and the H of the second carbon molecule. Put H-OH on the other C-OH and H on the other carbon molecule. Remove C-OH and H of the first and H of the second to form water.



Label atoms and other groups as needed.

3. Circle the atoms that form the C-O single bond that is in the new disaccharide. How many of these atoms are needed to form the new disaccharide?



What is the molecular formula for the new disaccharide? Circle the atoms in your structure.



What type of synthesis reaction is this synthesis?



What is the molecular formula for this reaction?



4. Label the synthesis reaction with the name of the reaction type in the box. How many of these atoms are needed to form the new disaccharide?



What is the molecular formula for this polymer unit?



Part B: Lipids

1. What are monosaccharides and disaccharides? How many of these atoms are needed to form the new disaccharide?



What is the molecular formula for glycerol?



What is the molecular formula for each of your fat units?

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Table of Contents Dehydration Synthesis Paper Activity

1. Understanding the eBook Dehydration Synthesis Paper Activity
 - The Rise of Digital Reading Dehydration Synthesis Paper Activity
 - Advantages of eBooks Over Traditional Books
2. Identifying Dehydration Synthesis Paper Activity
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Dehydration Synthesis Paper Activity
 - User-Friendly Interface
4. Exploring eBook Recommendations from Dehydration Synthesis Paper Activity
 - Personalized Recommendations
 - Dehydration Synthesis Paper Activity User Reviews and Ratings

- Dehydration Synthesis Paper Activity and Bestseller Lists
- 5. Accessing Dehydration Synthesis Paper Activity Free and Paid eBooks
 - Dehydration Synthesis Paper Activity Public Domain eBooks
 - Dehydration Synthesis Paper Activity eBook Subscription Services
 - Dehydration Synthesis Paper Activity Budget-Friendly Options
- 6. Navigating Dehydration Synthesis Paper Activity eBook Formats
 - ePub, PDF, MOBI, and More
 - Dehydration Synthesis Paper Activity Compatibility with Devices
 - Dehydration Synthesis Paper Activity Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Dehydration Synthesis Paper Activity
 - Highlighting and Note-Taking Dehydration Synthesis Paper Activity
 - Interactive Elements Dehydration Synthesis Paper Activity
- 8. Staying Engaged with Dehydration Synthesis Paper Activity
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Dehydration Synthesis Paper Activity
- 9. Balancing eBooks and Physical Books Dehydration Synthesis Paper Activity
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Dehydration Synthesis Paper Activity
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Dehydration Synthesis Paper Activity
 - Setting Reading Goals Dehydration Synthesis Paper Activity
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Dehydration Synthesis Paper Activity
 - Fact-Checking eBook Content of Dehydration Synthesis Paper Activity
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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