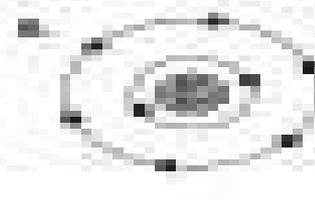
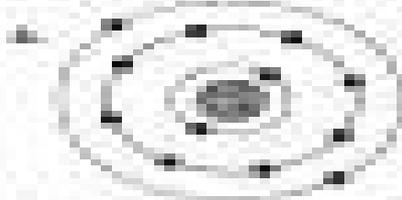


IPS Unit 9 – Chemical Bonds

Types of Bonds Worksheet



11) It was in 1913 discovered that:

- Atom shells (electrons) will glow if ionized
- Atom shells decrease with size of atom
- What within the shells of atoms is?
- What within the shells of atoms is?
- What explains the two shells of the hydrogen atom?
- What type of bond will form?

12) Explain why a substance's change in electrical charge is possible when a chemical change:

occurs and then describe the process of electron transfer, electron sharing or the gain/loss of electrons and how this affects the chemical change.

13) Complete the table for each compound by identifying its bonding type.

Compound	Bonding type	Electron arrangement
Sodium chloride (NaCl)	ionic bonding	electron transfer
Carbon dioxide (CO ₂)	covalent bonding	electron sharing
Water (H ₂ O)	covalent bonding	electron sharing
Iron (Fe)	metallic bonding	electron sea
Aluminum (Al)	metallic bonding	electron sea

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