

EASEL BONDING QUIZ

Multiple Choice: Select the most correct answer. (15 K)

1. An atom of argon rarely bonds to an atom of another element because an argon atom has:
A) 8 valence electrons B) 2 electrons in the 1st level
C) 3 energy levels D) 22 neutrons
2. The concept of multiple covalent bonds is used to explain the molecular formula of:
A) O₂ B) NaCl C) C₂H₄ D) F₂ E) NaOH
3. What type of bond is likely to form between the elements molybdenum, Mo and nitrogen, N based on electronegativities?
A) Non-polar covalent B) Polar covalent C) Ionic D) None of the above
4. In which pair do both compounds exhibit ionic bonding?
A) SO₂, HCl B) KNO₃, CH₄ C) NaF, K₂SO₄ D) KCl, CO₂ E) NaCl, H₂O
5. What type of bonding occurs between phosphorus and hydrogen within a molecule of PH₃?
A) ionic B) covalent C) hydrogen bonding D) metallic
6. What occurs in a triple covalent bond?
A) Two atoms share two electrons.
B) Two atoms share two pairs of electrons.
C) Two atoms share three pairs of electrons.
D) Two atoms share three electrons.
7. A molecule with a single covalent bond is _____.
A) CO₂ B) CO C) Cl₂ D) N₂
8. A solid has a very high melting point, is brittle, and has poor electrical conduction. This is a(n) _____ solid.
A) ionic B) molecular C) metallic
D) molecular and ionic E) metallic and ionic
9. Which of the following statements contrasting covalent bonds and ionic bonds is correct?
A) Covalent bonds usually involve 2 nonmetals and ionic bonds usually involve 2 metals.
B) Covalent bonds usually involve 2 metals and ionic bonds usually involve a metal and a nonmetal.
C) Covalent bonds usually involve a metal and a nonmetal and ionic bonds involve 2 nonmetals.
D) Covalent bonds usually involve 2 nonmetals and ionic bonds usually involve a metal and a nonmetal.

Section Quiz Introduction To Chemical Bonding Answers

Judd E. Hollander



Section Quiz Introduction To Chemical Bonding Answers:

Section Quiz Introduction To Chemical Bonding Answers Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Section Quiz Introduction To Chemical Bonding Answers**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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