

Worksheet 2: Binary Ionic Compounds (variable charge metals) - metals from groups 3 - 12 and 14 or 4A (Pb and Sn) - charge of anion is constant - use charge of anion to figure out charge of metal

1. Name the following ionic compounds:

|                            |                            |                         |
|----------------------------|----------------------------|-------------------------|
| a. $\text{SnO}_2$          | oxide = $\text{O}^{-2}$    | Tin (IV) oxide          |
| b. $\text{Mn}_2\text{O}_7$ | oxide = $\text{O}^{-2}$    | Manganese (VII) oxide   |
| c. $\text{FeN}$            | nitride = $\text{N}^{-3}$  | Iron (III) nitride      |
| d. $\text{Cu}_3\text{N}_2$ | nitride = $\text{N}^{-3}$  | Copper (II) nitride     |
| e. $\text{TiF}_3$          | fluoride = $\text{F}^{-1}$ | Titanium (III) fluoride |
| f. $\text{Cu}_2\text{S}$   | sulfide = $\text{S}^{-2}$  | Copper (I) sulfide      |
| g. $\text{Fe}_2\text{S}_3$ | sulfide = $\text{S}^{-2}$  | Iron (III) sulfide      |
| h. $\text{CuBr}$           | bromide = $\text{Br}^{-1}$ | Copper (I) bromide      |
| i. $\text{Co}_3\text{N}_2$ | nitride = $\text{N}^{-3}$  | Cobalt (II) nitride     |
| j. $\text{CoF}_2$          | fluoride = $\text{F}^{-1}$ | Cobalt (II) fluoride    |

2. Write the formula for the following ionic compounds:

|                           |                         |
|---------------------------|-------------------------|
| a. Copper (I) nitride     | $\text{Cu}_3\text{N}$   |
| b. Cobalt (I) fluoride    | $\text{CoF}$            |
| c. Titanium (IV) oxide    | $\text{TiO}_2$          |
| d. Iron (II) nitride      | $\text{Fe}_3\text{N}_2$ |
| e. Iron (III) chloride    | $\text{FeCl}_3$         |
| f. Copper (II) oxide      | $\text{CuO}$            |
| g. Rhodium (II) oxide     | $\text{RhO}$            |
| h. Tin (IV) sulfide       | $\text{SnS}_2$          |
| i. Manganese (IV) nitride | $\text{Mn}_3\text{N}_4$ |
| j. Copper (I) fluoride    | $\text{CuF}$            |
| k. Cobalt (II) sulfide    | $\text{CoS}$            |
| l. Iron (III) oxide       | $\text{Fe}_3\text{O}_2$ |

# Writing And Naming Binary Compounds Worksheet Answer Key

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