

PERIODIC TABLE **STRAND: UNDERSTANDING CHEMISTRY** **CHAPTER 11: PERIODICITY**

Section 11.1 *(continued)*

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

atomic mass	12.
atomic number	13.
electron configuration	14.
chemical name	15.
chemical symbol	16.

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|---|
| <p>_____ 17. A column in the periodic table</p> <p>_____ 18. A row in the periodic table</p> <p>_____ 19. Group 1 elements</p> <p>_____ 20. Elements that are shiny and conduct electricity</p> <p>_____ 21. Group 8 elements</p> | <p>a. metals</p> <p>b. gases</p> <p>c. period</p> <p>d. representative elements</p> <p>e. transition elements</p> |
|---|---|

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ **22.** There are two representative elements of platinum.
- _____ **23.** Most non-metal elements of the elements in the periodic table are metalloids.
- _____ **24.** Group 1A elements except for hydrogen are located on the left of metals.
- _____ **25.** Group 2A elements are the alkaline earth metals.
- _____ **26.** Group 7A elements are highly reactive nonmetals known as halogens.
- _____ **27.** Group 8A elements are inert nonmetallic elements known as noble gases.
- _____ **28.** Metalloids have properties of both metals and non-metallic elements.

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