

CHEMISTRY REVIEW: KEY CHAPTER 9 – CHEMICAL NAMES AND FORMULAS

- In a Stock name such as iron (II) sulfate, the roman numeral tells us **C**
 - How many atoms of Fe are in one formula unit
 - How many sulfate ions can be attached to the iron atom
 - The charge on each Fe ion
 - The total positive charge of the formula unit
- The result of changing a subscript in a correctly written chemical formula is to **c**
 - Change the number of moles represented in the formula
 - Change the charges on the other ions in the compound
 - Change the formula so that it no longer represents the compound
 - Have no effect on the formula
- Derive a generalization for when an acid name will end in the suffix **-ic** or **-ous**
If ending is -ic, the anion ended in -ate; if ending is -ous, the anion ended in -ite
- Derive a generalization for when an acid name will begin with the prefix **hydro-**
If prefix is hydro-, the anion ended in -ide. This is most likely a monatomic ion.
- Fill in the blanks in the table below

Compound Name	Formula
Aluminum sulfide	Al_2S_3
Aluminum sulfate	$Al_2(SO_4)_3$
Lead (II) chloride	$PbCl_2$
Ammonium phosphate	$(NH_4)_3PO_4$
Hydrobromic acid	HBr
Dinitrogen pentoxide	N_2O_5
Copper (II) carbonate	$CuCO_3$
Sodium sulfide	Na_2S
Ammonium phosphate	$(NH_4)_3PO_4$
Tin (IV) sulfide	$Sn(SO_4)_2$
Nitrous acid	HNO_2
Magnesium perchlorate	$Mg(ClO_4)_2$
Iron (II) nitrate	$Fe(NO_3)_2$
Iron (III) nitrate	$Fe(NO_3)_3$
Cobalt (II) oxide	CoO
Sulfur trioxide	SO_3
Boron pentafluoride	BF_5
Tetraphosphorus octasulfide	P_4S_8
Tetraoxide nonoxide	TeO_9

- Name the following compound and identify as molecular, ionic or acid
 - $Ca(HCO_3)_2$ calcium bicarbonate - **I**
 - $Fe(ClO_2)_3$ iron (III) chlorite - **I**
 - $Fe(ClO_4)_2$ iron (II) chlorate - **I**
 - N_2O_5 dinitrogen pentoxide - **M**

Chapter 9 Chemical Names And Formulas Answers

James R. Braun, Sherman



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Understanding Chemistry Fred M. Dewey,1994 ... *Chemistry* Jay Ellis Stannard,1927 **Chemistry** Henry Dorin,1987 **Student's Guide to Introduction to Chemical Principles by Edward I. Peters, 2d Ed** Peter P. Berlow,1978 **Essentials of Chemistry** Dennis D. Staley, Antony C. Wilbraham,1984 **Essential Concepts of Chemistry Study Guide** James R. Braun, Sherman,1999 10 in One Study Package for CBSE Science Class 10 with 3 Sample Papers & 16 Chapter Tests ebook Disha Experts,2017-09-01 These books contain Access Codes along with instructions to access the Online Material In case you face any difficulty write to us at ebooks support aietco in 10 in ONE CBSE Study Package Science class 10 with 3 Sample Papers is another innovative initiative from Disha Publication This book provides the excellent approach to Master the subject The book has 10 key ingredients that will help you achieve success 1 Chapter Utility Score CUS 2 Exhaustive Theory with Concept Maps 3 Text Book exercises 4 VSA SA LA Questions 5 Past year questions including 2017 Solved papers 6 HOTS Value based Exemplar 7 Past NTSE Exemplar MCQ s 8 16 Chapter Tests ebooks 9 Important Formulas Terms Definitions 10 3 Sample Papers with detailed solutions *The Digester, Over the Spillway*,1976 **Structure and Change** Alan C. Blizard, Richardson, R. A. (Robert A.), Humphreys, D. A. (David A.),1981 *Essential Concepts of Chemistry* Sharon J. Sherman,1998-09 **Pharmacology for the Surgical Technologist - E-Book** Katherine Snyder, Chris Keegan,2016-01-19 Learn with the most trusted pharmacology text written specifically for surgical technologists Pharmacology for the Surgical Technologist 4th Edition ensures that as an integral member of the operating room team you have an in depth understanding of surgical medications It covers everything from basic conversions terminology and calculations to anesthesia and medications used for perioperative care This edition adds coverage of next generation antibiotics new technology in orthopedics advances in cataract surgery and more plus a full color design and expanded art program An Evolve companion website lets you practice drug calculations Written by experienced Surgical Technology educators Katherine Snyder and Chris Keegan this book covers all areas of pharmacology that are designated in the Core Curriculum for Surgical Technology 6th Edition Coverage of pharmacology meets the needs of the Surgical Technologist and includes all areas designated in the Core Curriculum for Surgical Technology 6th Edition Chapter study questions help you measure your knowledge and apply it to practice and serve as a review tool for classroom and certification exams Clinical features include Tech Tips from experts Caution boxes with drug alerts to help prevent potential patient harm and Notes simplifying difficult concepts Insight boxes provide in depth cutting edge information about a specific product procedure or process Concise three part organization makes it easier to understand 1 the foundations of pharmacology mathematics and drug administration 2 applications of pharmacology to the surgical environment and 3 preoperative medications types of anesthesia and emergency situations Evolve companion website includes exercises allowing you to practice math calculations as well as drug monographs from Mosby's Essential Drugs for Surgical Technology

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