

Balancing and Reactions Test Unit 2 SCH3U

Name: _____

 K T C A **Assessment of Learning (Product)**
15 20 5 4

Multiple Choice: Circle the most correct answer. (10K)

1. Propane gas, C_3H_8 , burns in excess oxygen gas. When the equation for this reaction is correctly balanced and all coefficients are reduced to their lowest whole-number terms, the coefficient for H_2O is...

- A) 1 B) 2 C) 3 D) 4 E) 5 f) 6

2. State the type of reaction: $2 Na + 2 H_2O \rightarrow 2 NaOH + H_2$

- A) synthesis B) decomposition C) single displacement
-
- D) double displacement E) combustion F) no reaction

3. Which of the following types of chemical reactions produces water and an ionic compound?

- A) single displacement B) neutralization C) decomposition D) synthesis

4. State the type of reaction: $Ca(NO_3)_2 + 2 HCl \rightarrow CaCl_2 + 2 HNO_3$

- A) synthesis B) decomposition C) single displacement
-
- D) double displacement E) combustion F) no reaction

5. If chlorine gas is produced by single displacement, the other halogen in the reaction must be

- A) bromine B) iodine C) astatine D) fluorine

6. In what kind of reaction do two or more substances combine to form a new compound?

- A) decomposition reaction C) double-displacement reaction
-
- B) single displacement D) synthesis reaction

7. A group whose atoms (M) undergo the reaction:



- A) group 1 B) group 2 C) group 13 D) group 14 E) group 15

8. When heated, a metal carbonate decomposes into a metal oxide and

- A) carbon. B) carbon dioxide C) oxygen. D) hydrogen.

9. Oxides of active metals, such as CaO , react with water to produce

- A) metal carbonates. B) metal hydrides. C) acids. D) metal hydroxides.

10. Which of the following statements is true about single-displacement reactions?

- A) They are restricted to metals. C) Two reactants produce two products.
-
- B) They involve a single product. D) Any metal replaces any other metal.

Chemical Equations And Reactions Test Answers

Albert A Gayle



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