

Solution Terminology: Multiple Choice

1. A substance whose water solution does not conduct an electric current is a(n)
A) solute B) solution C) electrolyte D) immiscible E) concentrated
F) soluble G) solvent H) unsaturated I) dilute J) non-electrolyte
K) supersaturated L) miscible

ANS: J

1. Liquids that dissolve freely in one another in any proportion are called
A) solute B) solution C) electrolyte D) immiscible E) concentrated
F) soluble G) solvent H) unsaturated I) dilute J) non-electrolyte
K) supersaturated L) miscible

ANS: L

1. When a lot of solid dissolves, we refer to this solid as being
A) solute B) solution C) electrolyte D) immiscible E) concentrated
F) soluble G) solvent H) unsaturated I) dilute J) non-electrolyte
K) supersaturated L) miscible

ANS: F

1. When a solid dissolves in a liquid, the solid is called the
A) solute B) solution C) electrolyte D) immiscible E) concentrated
F) soluble G) solvent H) unsaturated I) dilute J) non-electrolyte
K) supersaturated L) miscible

ANS: A

1. The dissolving medium in a solution is called
A) solute B) solution C) electrolyte D) immiscible E) concentrated
F) soluble G) solvent H) unsaturated I) dilute J) non-electrolyte
K) supersaturated L) miscible

ANS: G

1. A substance that dissolves in water to give a solution that conducts electric current is called
A) solute B) solution C) electrolyte D) immiscible E) concentrated
F) soluble G) solvent H) unsaturated I) dilute J) non-electrolyte
K) supersaturated L) miscible

Multiple Choice Solution Chemistry 11 Questions

Robert R. Redfield



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