



# COVALENT BONDING

## Practice Problems

Do your answers to the following problems.

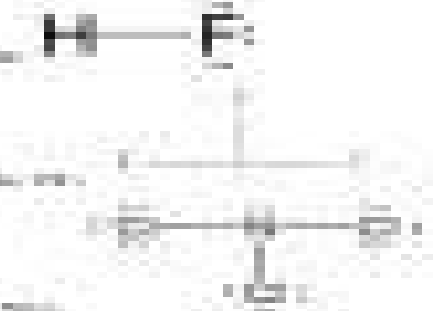
### SECTION 8.1 MOLECULAR COMPOUNDS

- Classify each of the following as either an ionic or a molecular compound.
  - $\text{NaCl}$
  - $\text{H}_2\text{O}$
  - $\text{HCl}$
  - $\text{KNO}_3$
  - $\text{C}_2\text{H}_6$
  - $\text{H}_2\text{SO}_4$
  - $\text{CaCl}_2$
  - $\text{C}_2\text{H}_5\text{OH}$
  - $\text{CO}_2$
  - $\text{H}_2\text{O}_2$
  - $\text{CH}_4$
  - $\text{H}_2\text{S}$

- Classify each of the following as either an ionic or a molecular compound. Write the name of the compound.
- Write the formula for each of the following molecular compounds. Write the name of the compound.
- Write the formula for each of the following ionic compounds. Write the name of the compound.

### SECTION 8.2 THE NATURE OF COVALENT BONDING

- Show covalent bonding in hydrogen fluoride,  $\text{H}-\text{F}$ .
- Show covalent bonding in methane,  $\text{CH}_4$ .
- Show covalent bonding in carbon dioxide,  $\text{CO}_2$ .



- Show covalent bonding in acetylene,  $\text{H}-\text{C}\equiv\text{C}-\text{H}$ .
- Write the Lewis structure for each of the following molecular compounds. Show covalent bonding.



### SECTION 8.3 BONDING IN THE LHS

- Write the Lewis structure for each of the following molecular compounds. Show covalent bonding.
- Write the Lewis structure for each of the following ionic compounds. Show covalent bonding.

# Chapter 8 Covalent Bonding Work Answers Pearson

**M Lipman**



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## Enjoying the Beat of Expression: An Emotional Symphony within **Chapter 8 Covalent Bonding Work Answers Pearson**

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### **Table of Contents Chapter 8 Covalent Bonding Work Answers Pearson**

1. Understanding the eBook Chapter 8 Covalent Bonding Work Answers Pearson
  - The Rise of Digital Reading Chapter 8 Covalent Bonding Work Answers Pearson
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 8 Covalent Bonding Work Answers Pearson
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 8 Covalent Bonding Work Answers Pearson
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 8 Covalent Bonding Work Answers Pearson

- Personalized Recommendations
  - Chapter 8 Covalent Bonding Work Answers Pearson User Reviews and Ratings
  - Chapter 8 Covalent Bonding Work Answers Pearson and Bestseller Lists
5. Accessing Chapter 8 Covalent Bonding Work Answers Pearson Free and Paid eBooks
    - Chapter 8 Covalent Bonding Work Answers Pearson Public Domain eBooks
    - Chapter 8 Covalent Bonding Work Answers Pearson eBook Subscription Services
    - Chapter 8 Covalent Bonding Work Answers Pearson Budget-Friendly Options
  6. Navigating Chapter 8 Covalent Bonding Work Answers Pearson eBook Formats
    - ePub, PDF, MOBI, and More
    - Chapter 8 Covalent Bonding Work Answers Pearson Compatibility with Devices
    - Chapter 8 Covalent Bonding Work Answers Pearson Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Chapter 8 Covalent Bonding Work Answers Pearson
    - Highlighting and Note-Taking Chapter 8 Covalent Bonding Work Answers Pearson
    - Interactive Elements Chapter 8 Covalent Bonding Work Answers Pearson
  8. Staying Engaged with Chapter 8 Covalent Bonding Work Answers Pearson
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Chapter 8 Covalent Bonding Work Answers Pearson
  9. Balancing eBooks and Physical Books Chapter 8 Covalent Bonding Work Answers Pearson
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Chapter 8 Covalent Bonding Work Answers Pearson
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Chapter 8 Covalent Bonding Work Answers Pearson
    - Setting Reading Goals Chapter 8 Covalent Bonding Work Answers Pearson
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Chapter 8 Covalent Bonding Work Answers Pearson

- Fact-Checking eBook Content of Chapter 8 Covalent Bonding Work Answers Pearson
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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