

Physical Science Chapter 66 (Properties of Atoms and the Periodic Table) Worksheet 1

Name: _____

Period: _____

Use your notebook to define the following terms:

Atom: _____

Atomic number: _____

Average atomic mass: _____

Atomic mass: _____

Electron: _____

Electron cloud: _____

Electron dot diagram: _____

Group: _____

Isotope: _____

Mass number: _____

Neutron: _____

Nucleus: _____

Period: _____

Periodic table: _____

Proton: _____

Quark: _____

Multiple choice (Choose the best answer):

- 1) In which state of matter are most of the elements to the left of the stair-step line in the periodic table?
 - a) Gas
 - b) Liquid
 - c) Plasma
 - d) Solid

- 2) Which is a name for a pattern that repeats?
 - a) Isotope
 - b) Mendeleev
 - c) Periodic
 - d) Transition

- 3) Which of the following methods would be most useful for describing atoms?
 - a) Visualizing with microscopes
 - b) Observing in a test tube
 - c) Constructing a model
 - d) Touching with bare hands

- 4) An atom containing six positive charges and six negative charges has _____ charge.
 - a) a positive
 - b) a negative
 - c) a strong negative
 - d) no net

Chapter Test Properties Of Atoms The Periodic Table Answer Key

M Tight



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