

CHEMICAL BONDING POGIL

Activity 2 – What's in a name?

Objectives:

- Identify some simple rules about nomenclature (naming).

The Model:

Examine the table below, and answer the following questions.

Table 2

Cation	Anion	Chemical Formula	Compound Name
Na^+	Cl^-	NaCl	sodium chloride
Ca^{2+}	O^{2-}	CaO	calcium oxide
Zn^{2+}	Cl^-	ZnCl_2	zinc chloride
Li^+	S^{2-}	Li_2S	lithium sulfide
K^+	N^{3-}	K_3N	potassium nitride

Reviewing the Model

- Are ALL cations positive ions or negative ions? *positive*
- Are ALL anions positive ions or negative ions? *negative*
- What is the name of the compound formed by the combination of Li^+ and S^{2-} ions? *Lithium sulfide*

Exploring the Model

- When the name of an ionic compound is given, which ion is stated first?
Cation
- Compare the first part of the compound name to the name of the element from the periodic table. How does the name of the cation correspond to the name of the element?
- Compare the second part of the compound name to the name of the element from the periodic table. How does the name of the anion correspond to the name of the element?
- From what part of the periodic table do the cations in the Model come (metals or nonmetals)?
Alkali and alkaline earth
- From what part of the periodic table do the anions in the Model come?
4

Chemical Bonding Pogil Answers Key

R Sandford



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Table of Contents Chemical Bonding Pogil Answers Key

1. Understanding the eBook Chemical Bonding Pogil Answers Key
 - The Rise of Digital Reading Chemical Bonding Pogil Answers Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Bonding Pogil Answers Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Bonding Pogil Answers Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Bonding Pogil Answers Key
 - Personalized Recommendations
 - Chemical Bonding Pogil Answers Key User Reviews and Ratings
 - Chemical Bonding Pogil Answers Key and Bestseller Lists

5. Accessing Chemical Bonding Pogil Answers Key Free and Paid eBooks
 - Chemical Bonding Pogil Answers Key Public Domain eBooks
 - Chemical Bonding Pogil Answers Key eBook Subscription Services
 - Chemical Bonding Pogil Answers Key Budget-Friendly Options
6. Navigating Chemical Bonding Pogil Answers Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Bonding Pogil Answers Key Compatibility with Devices
 - Chemical Bonding Pogil Answers Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Bonding Pogil Answers Key
 - Highlighting and Note-Taking Chemical Bonding Pogil Answers Key
 - Interactive Elements Chemical Bonding Pogil Answers Key
8. Staying Engaged with Chemical Bonding Pogil Answers Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Bonding Pogil Answers Key
9. Balancing eBooks and Physical Books Chemical Bonding Pogil Answers Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Bonding Pogil Answers Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemical Bonding Pogil Answers Key
 - Setting Reading Goals Chemical Bonding Pogil Answers Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemical Bonding Pogil Answers Key
 - Fact-Checking eBook Content of Chemical Bonding Pogil Answers Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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