

MCAT PRACTICE TEST 92 QUESTIONS AND ANSWERS 2024

Consider a biochemical reaction A to B, which is catalyzed by A-B dehydrogenase. Which of the following statements is true?

- A. The reaction will proceed until the enzyme concentration decreases.
 - B. The reaction will be more favorable at 0 degrees C.
 - C. A component of the enzyme transferred from A to B.
 - D. The free energy change (ΔG) of the catalyzed reaction is the same as for the uncatalyzed reaction.
- D.
- In an enzyme-catalyzed reaction, the rate of a reaction is increased by a decrease in the activation energy. Furthermore, enzymes are not changed or consumed during the course of the reaction. Also, the overall free energy change of the reaction, ΔG , remains unchanged in the presence of an enzyme.

Which of the following statements about enzyme kinetics is false?

- A. An increase in the substrate concentration (at constant enzyme concentration) leads to proportional increases in the rate of the reaction.
- B. Most enzymes operating in the human body work best at a temperature of 37 degree C.
- C. An enzyme-substrate complex can either form a product or dissociate back into the enzyme and substrate.
- D. Maximal activity of many human enzymes occurs around pH 7.2.

A.

-Most enzymes in the human body operate a maximal activity around a temperature of 37 degrees C and a pH of 7.2, which is the pH of most body fluids. In addition, as characterized by the Michaelis-Menten model, enzymes form an enzyme-substrate complex, which can either dissociate back into the enzyme and substrate or proceed to form a product. We can eliminate B, C, and D. An increase in the substrate concentration, while maintaining a constant enzyme concentration, leads to a proportional increase in the rate of the reaction only initially. However, once most of the active sites are occupied, the reaction rate levels off, regardless of further increases in substrate concentration. At high concentrations of substrate, the reaction rate approaches its maximal velocity and is no longer changed by further increases in substrate concentration. Therefore statement A is not entirely true.

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Table of Contents Practice Mcat Questions And Answers

1. Understanding the eBook Practice Mcat Questions And Answers
 - The Rise of Digital Reading Practice Mcat Questions And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Practice Mcat Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Practice Mcat Questions And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Practice Mcat Questions And Answers
 - Personalized Recommendations
 - Practice Mcat Questions And Answers User Reviews and Ratings
 - Practice Mcat Questions And Answers and Bestseller Lists
5. Accessing Practice Mcat Questions And Answers Free and Paid eBooks
 - Practice Mcat Questions And Answers Public Domain eBooks

- Practice Mcat Questions And Answers eBook Subscription Services
- Practice Mcat Questions And Answers Budget-Friendly Options
- 6. Navigating Practice Mcat Questions And Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Practice Mcat Questions And Answers Compatibility with Devices
 - Practice Mcat Questions And Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Practice Mcat Questions And Answers
 - Highlighting and Note-Taking Practice Mcat Questions And Answers
 - Interactive Elements Practice Mcat Questions And Answers
- 8. Staying Engaged with Practice Mcat Questions And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Practice Mcat Questions And Answers
- 9. Balancing eBooks and Physical Books Practice Mcat Questions And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Practice Mcat Questions And Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Practice Mcat Questions And Answers
 - Setting Reading Goals Practice Mcat Questions And Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Practice Mcat Questions And Answers
 - Fact-Checking eBook Content of Practice Mcat Questions And Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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