

Name :

Date :

# Chemistry Atomic Structure Worksheet

① For the atomic structure shown below right, answer the following questions.

a) Which element is the atom shown above? \_\_\_\_\_

b) What is the chemical symbol of the element? \_\_\_\_\_

c) What is the mass of the atom shown above? \_\_\_\_\_

d) How many valence electrons are shown in the atom above? \_\_\_\_\_



② Which subatomic particle did J.J. Thomson discover in order to create his plum pudding model of the atom? \_\_\_\_\_

③ The nucleus was discovered by which scientist:

a) Thomson

b) Dalton

c) Rutherford

d) Democritus

④ What is the mass of Carbon-14: \_\_\_\_\_

⑤ What is the number of protons in Carbon-14: \_\_\_\_\_

⑥ What is the number of neutrons in Carbon-14: \_\_\_\_\_

⑦ What is the number of protons in the radioactive Uranium-238: \_\_\_\_\_

⑧ What is the number of neutrons in Sulfur-33: \_\_\_\_\_

⑨ Neutral atoms of the same element can differ in

a) Mass number

b) Atomic number

c) Number of protons and electrons

⑩ The basic principles of atomic theory were first conceived by

a) Avogadro

b) Bohr

c) Dalton

# Pearson Chemistry Atomic Structure Test Answers

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## **Table of Contents Pearson Chemistry Atomic Structure Test Answers**

1. Understanding the eBook Pearson Chemistry Atomic Structure Test Answers
  - The Rise of Digital Reading Pearson Chemistry Atomic Structure Test Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Chemistry Atomic Structure Test Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pearson Chemistry Atomic Structure Test Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Chemistry Atomic Structure Test Answers
  - Personalized Recommendations
  - Pearson Chemistry Atomic Structure Test Answers User Reviews and Ratings
  - Pearson Chemistry Atomic Structure Test Answers and Bestseller Lists

5. Accessing Pearson Chemistry Atomic Structure Test Answers Free and Paid eBooks
  - Pearson Chemistry Atomic Structure Test Answers Public Domain eBooks
  - Pearson Chemistry Atomic Structure Test Answers eBook Subscription Services
  - Pearson Chemistry Atomic Structure Test Answers Budget-Friendly Options
6. Navigating Pearson Chemistry Atomic Structure Test Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Pearson Chemistry Atomic Structure Test Answers Compatibility with Devices
  - Pearson Chemistry Atomic Structure Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Pearson Chemistry Atomic Structure Test Answers
  - Highlighting and Note-Taking Pearson Chemistry Atomic Structure Test Answers
  - Interactive Elements Pearson Chemistry Atomic Structure Test Answers
8. Staying Engaged with Pearson Chemistry Atomic Structure Test Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pearson Chemistry Atomic Structure Test Answers
9. Balancing eBooks and Physical Books Pearson Chemistry Atomic Structure Test Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pearson Chemistry Atomic Structure Test Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Pearson Chemistry Atomic Structure Test Answers
  - Setting Reading Goals Pearson Chemistry Atomic Structure Test Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pearson Chemistry Atomic Structure Test Answers
  - Fact-Checking eBook Content of Pearson Chemistry Atomic Structure Test Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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