

# Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

Chemistry

Physical Science

## SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

What is a homogeneous mixture? Provide two examples.

Describe the characteristics of a saturated solution.

1. Define the terms solvent and solute as they apply to a solution.

2. Define solubility, and how is it measured?

3. Distinguish between concentrated and dilute.

4. List the properties of bases and acids.

5. Explain the significance of the pH scale.

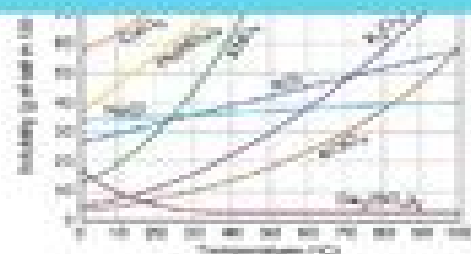
6. How does the concentration of an acid affect its pH?

7. Explain the relationship between pH and pOH.

8. Explain the relationship between pH and pKa.

9. Explain the relationship between pH and pKb.

10. Explain the relationship between pH and pKsp.



11. Describe how you would use the graph to determine the amount of  $\text{NaCl}$  that can be dissolved in 100 grams of water at  $80^\circ\text{C}$ .

12. Which salt has the highest solubility at  $80^\circ\text{C}$ ?



ANSWER KEY INCLUDED

# Acids And Bases Study Guide Key

**Steven S. Zumdahl, Paul B. Kelter**



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**Study Guide with Answers to Selected Problems** H. Stephen Stoker, Danny V. White, 2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises

**Study Guide with Selected Solutions** Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

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*Study Guide for Introductory Chemistry, a Foundation, Introductory Chemistry, Basic Chemistry* Iris Stovall, Steven S. Zumdahl, 1996 *General Chemistry Study Guide Sixth Edition* Darrell D. Ebbing, 1999 [Chemical Interactions](#), 2005

*Prentice Hall Chemistry*, 2000 *Organic Chemistry* Michael B. Smith, 2016-03-09 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts *Organic Chemistry An Acid Base Approach* provides a framework for understanding the subject that goes beyond mere memorization. Using several techniques to develop a relational understanding it helps students fully grasp the essential concepts at the root of organic chemistry. This new edition was rewritten largely with the feedback of students in mind and is also based on the author's classroom experiences using the first edition. Highlights of the Second Edition include reorganized chapters that improve the presentation of material, coverage of new topics such as green chemistry, adding photographs to the lectures to illustrate and emphasize important concepts, a downloadable solutions manual. The second edition of *Organic Chemistry An Acid Base Approach* constitutes a significant improvement upon a unique introductory technique to organic chemistry. The reactions and mechanisms it covers are the most fundamental concepts in organic chemistry that are applied to industry, biological chemistry, biochemistry, molecular biology and pharmacy. Using an illustrated conceptual approach rather than presenting sets of principles and theories to memorize it gives students a more concrete understanding of the material.

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