

### 7B-3 : Entropy Change of an Isobaric Process

6 pts

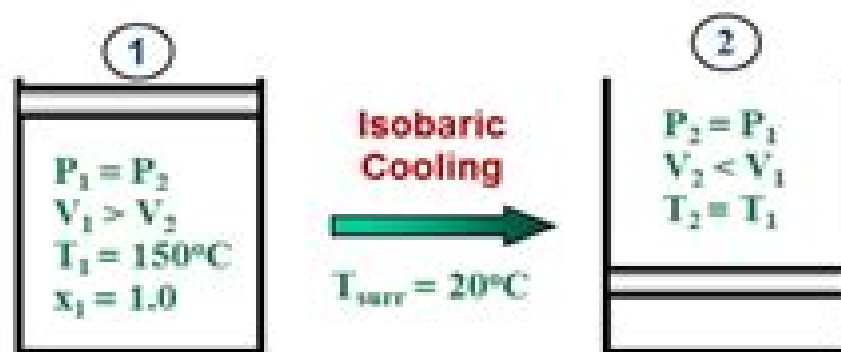
A piston-and-cylinder device with a free-floating piston contains 2.6 kg of saturated water vapor at 150°C. The water loses heat to the surroundings until the cylinder contains saturated liquid water.

The surroundings are at 20°C. Calculate... a.)  $\Delta S_{\text{water}}$ . b.)  $\Delta S_{\text{surroundings}}$ . c.)  $\Delta S_{\text{universe}}$

**Read :** Calculating  $\Delta S$  for the water in the cylinder is straightforward. The key to calculating  $\Delta S_{\text{surr}}$  is the fact that the surroundings behave as a thermal reservoir. The temperature of the surroundings does not change and there are no irreversibilities in the surroundings that are associated with the process. The key to calculating  $\Delta S_{\text{univ}}$  is the fact that the universe is made up of the combination of the system and the surroundings. Consequently,  $\Delta S_{\text{univ}} = \Delta S_{\text{sys}} + \Delta S_{\text{surr}}$ .

|               |                   |      |           |              |                          |     |      |
|---------------|-------------------|------|-----------|--------------|--------------------------|-----|------|
| <b>Given:</b> | $m$               | 2.60 | kg        | <b>Find:</b> | $\Delta S_{\text{sys}}$  | ??? | kJ/K |
|               | $T_1$             | 150  | °C        |              | $\Delta S_{\text{surr}}$ | ??? | kJ/K |
|               | $x_1$             | 1    | kg vap/kg |              | $\Delta S_{\text{univ}}$ | ??? | kJ/K |
|               | $T_{\text{surr}}$ | 20   | °C        |              |                          |     |      |
|               | $P_1 = P_2$       |      |           |              |                          |     |      |
|               | $x_2$             | 0    | kg vap/kg |              |                          |     |      |

**Diagram:**



- Assumptions:**
1. Changes in kinetic and potential energies are negligible.
  2. Boundary work is the only form of work that crosses the system boundary.
  3. The surroundings behave as a thermal reservoir.

# Entropy Problems And Solutions

**M Mosston**



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