

SALT ANALYSIS

1) For $MgSO_4$ - magnesium Sulfate :-

Aim: To analyze the given salt for an anion and one cation present in it.

Material Required: Boiling Tube, Test Tubes, Test Tube Holder, Test Tube Stand, delivery tube, Corks, Filter paper, Test paper.

Procedure:-

Sr.	Test	Observation	Inference
i)	Colour	White	
ii)	Odour		
iii)	Flame test	Greenish white	
iv)	Wet H_2SO_4 test (Salt + conc. H_2SO_4)	No ppt. form.	
v)	Char. H_2SO_4 test (Salt + conc. H_2SO_4)	No.	

ii) For $Fe^{2+}Cl^-$, Ferrous chloride :-

Aim: To analyze the given salt for an anion and one cation present in it.

Material Required: Boiling Tube, Test Tube, Test Tube Holder, Test Tube Stand, delivery tube, Corks, Filter paper, Test paper.

Procedure:-

Sr.	Test	Observation	Inference
		Brown reddish	Fe^{2+} may be present
		Colourless	NO_3^- , S^{2-} , CO_3^{2-} are absent
			Ca^{2+} , S^{2-} , SO_4^{2-} , NO_3^- are absent
			Cl^- may be present
			Cl^- confirmed

Class 12th

Chemistry Practical File

Experiment - SALT ANALYSIS

5 most Important Salt

Handwritten PDF Notes

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web jul 7 2023 ans an ideal fluid is one that cannot be compressed and has no viscosity b describe capillary rise ans capillary rise is the rise of a liquid surface inside a tiny tube compared to the surrounding liquid's general level while the tube is held vertically in the liquid c define froude's number ans 1

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