

Reference Tables for Physical Setting/CHEMISTRY

2002 Edition

Table A
Standard Temperature and Pressure

Name	Value	Unit
Standard Pressure	101.3 kPa 1 atm	kilopascal atmosphere
Standard Temperature	273 K 0°C	kelvin degree Celsius

Table B
Physical Constants for Water

Heat of Fusion	334 J/g
Heat of Vaporization	2260 J/g
Specific Heat Capacity of H ₂ O (l)	4.18 J/g•°C

Table C
Selected Prefixes

Factor	Prefix	Symbol
10 ³	kilo-	k
10 ⁻¹	deci-	d
10 ⁻²	centi-	c
10 ⁻³	milli-	m
10 ⁻⁶	micro-	μ
10 ⁻⁹	nano-	n
10 ⁻¹²	pico-	p

Table D
Selected Units

Symbol	Name	Quantity
m	meter	length
g	gram	mass
Pa	pascal	pressure
K	kelvin	temperature
mol	mole	amount of substance
J	joule	energy, work, quantity of heat
s	second	time
L	liter	volume
ppm	part per million	concentration
M	molarity	solution concentration

Table E
Selected Polyatomic Ions

H ₃ O ⁺	hydronium	CrO ₄ ²⁻	chromate
Hg ₂ ²⁺	mercurous (II)	Cr ₂ O ₇ ²⁻	dichromate
NH ₄ ⁺	ammonium	MnO ₄ ⁻	permanganate
C ₂ H ₃ O ₂ ⁻ CH ₃ COO ⁻	acetate	NO ₂ ⁻	nitrite
		NO ₃ ⁻	nitrate
CN ⁻	cyanide	O ₂ ²⁻	peroxide
CO ₃ ²⁻	carbonate	OH ⁻	hydroxide
HCO ₃ ⁻	hydrogen carbonate	PO ₄ ³⁻	phosphate
C ₂ O ₄ ²⁻	oxalate	SCN ⁻	thiocyanate
ClO ⁻	hypochlorite	SO ₃ ²⁻	sulfite
ClO ₂ ⁻	chlorite	SO ₄ ²⁻	sulfate
ClO ₃ ⁻	chlorate	H ₂ SO ₄	hydrogen sulfate
ClO ₄ ⁻	perchlorate	S ₂ O ₃ ²⁻	thiosulfate

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William Docekal, Ronald Pasto



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