



## Lab Data



Verify your calculation. In your set-up, do your units cancel?

Mass of magnesium (g)	.015
Moles of magnesium (mol)	.00062
Temperature of water (°C)	23.0
Temperature of water (K)	296.15
Vapor pressure of water (mmHg)	21.1
Barometric pressure (mmHg)	778.0

## Observations

I observe that there are bubbles that are visible.

Volume of hydrogen gas collected (mL)	15.00
Volume of hydrogen gas collected (L)	0.01500
Pressure of hydrogen gas (mmHg)	756.9
Pressure of hydrogen gas (atm)	0.9959
Moles of hydrogen gas produced (mol)	.00690
Ideal gas constant (L · atm · mol <sup>-1</sup> · K <sup>-1</sup> )	0.080

APR  
15

# Introduction To Gas Law Lab Answer Key

**Jeffrey Stanton**



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Mining Science and Technology Yuehan Wang, Shirong Ge, Guangli Guo, 2004-09-15 Jointly sponsored by the China University of Mining and Technology and the University of Nottingham UK a total of 187 papers have been included in the proceedings of which fifty two are contributed by authors outside of China Scholars and experts from both China and abroad discuss and exchange information on the latest developments in mining sc

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**An Introductory Guide to EC Competition Law and Practice** Valentine Korah, 1994

**Paperbound Books in Print**, 1969

*The Book of Apple Computer Software*, 1981 Kept up to date by quarterly supplements

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*Biology/science Materials* Carolina Biological Supply Company, 1991

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