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 $2 \text{ } 3 \text{ ca oh} + 2 \text{ } 1 \text{ al}_2 \text{ so}_4 \rightarrow 3 \text{ } 3 \text{ caso}_4 + 2 \text{ } 1 \text{ al oh}$ type of reaction double displacement
 $3 \text{ } 3 \text{ mg} + 1 \text{ } 2 \text{ o}_3 \rightarrow 2 \text{ } 3 \text{ mgo}$ type of reaction single displacement
 $4 \text{ } 1 \text{ c}_2\text{h}_4 + 3 \text{ } 1 \text{ o}_2 \rightarrow 2 \text{ } 2 \text{ co}_2$

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