

Chemistry Chapter 8 Exam

Multiple Choice

1. Which of the following is not a Lewis acid?

- _____ a. CH_3COOH
- _____ b. Lewis Acids
- _____ c. Carbonyl Compounds
- _____ d. Alcohols
- _____ e. Metal Ions
- _____ f. Simple Molecules
- _____ g. Molecule Molecules
- _____ h. Lewis Acids
- _____ i. Hydrogen Ions
- _____ j. None of the above

- _____ k. bonded to electronegative atoms that are more electronegative than carbon
- _____ l. single group of molecules bonded to one
- _____ m. the energy released when one mole of an ionic crystalline compound dissolves in water
- _____ n. coordination bonds are part of a molecule's chemical composition
- _____ o. the energy required to break a chemical bond and form atomic ions
- _____ p. structural groups of atoms that are held together by covalent bonds, usually form the basis of chemical composition
- _____ q. chemical bonding that results from the electrical attraction between positive and negative charges
- _____ r. mutual electrical attraction between protons and electrons in an atom or between atoms, molecules, ions, or groups

Multiple Choice

2. The charge on a bond is _____

- _____ a. always positive
- _____ b. always negative
- _____ c. either positive or negative
- _____ d. neither

3. In many compounds, atoms of more electronegative elements form bonds with the atoms of less electronegative elements. Which of the following is not true?

- _____ a. The bond is polar.
- _____ b. The bond is ionic.
- _____ c. The bond is covalent.
- _____ d. The bond is nonpolar.

4. Which of the following is not a Lewis acid? _____

Modern Chemistry Chapter 6 Test Answers

H Kauffman



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