12  a) Describe the construction and working of Li-ion battery. What are the major advantages of it?  (10)

    b) Calculate the EMF of the cell at $25°C$:  (4)

    $Al / Al^{3+}(0.1M) // Sn^{4+}(0.1M) / Sn^{2+}(0.01M)$

    $E^0_{Al^{3+}/Al} = -1.66 V$,   $E^0_{Sn^{4+}/Sn^{2+}} = 0.15 V$

### Module-II

13  a) Define chemical shift in NMR and explain the factors affecting chemical shift with examples.  (8)

    b) Calculate the force constant of H-F molecule that is showing IR absorption signal at $4000 cm^{-1}$.  By what factor do you expect this frequency to shift if Deuterium is substituted for Hydrogen in this molecule? Given that atomic masses of H and F are 1u and 19 u, respectively.  (6)

14  a) How many vibrational modes are possible for the molecules CO, NO, $CO_2$ and $H_2O$? Draw the vibrational modes of $CO_2$ and $H_2O$ and explain their IR active modes.  (8)

    b) Each compound gives only one signal in its 1H-NMR spectrum. Propose a structural formula for each. a) $C_8H_{18}$ and  b) $C_8H_{18}O$  (6)

### Module-III

15  a) Describe the instrumentation, principle and working of SEM with the help of a labelled diagram. Give any two applications.  (10)

    b) Explain the visualization techniques used in thin layer chromatography.  (4)

16  a) Describe the principle, instrumentation, procedure and applications of HPLC.  (10)

    b) How TGA is used to analyze thermal stability of polymers?  (4)

### Module-IV

17  a) Discuss the conformations in butane by depicting the Newman formula about C2-C3 bond of all the conformers. Also draw the energy level diagram with dihedral angle.  (10)

    b) How is ABS synthesized? Mention some applications.  (4)

18  a) Explain the rules for assigning **R-S** configuration and determine the **R-S** configuration of all the asymmetric carbon atoms in the molecule after writing its Fischer projection formula.  (8)

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**Table of Contents Engineering Chemistry Paper Solved 2010**

1. Understanding the eBook Engineering Chemistry Paper Solved 2010
    - The Rise of Digital Reading Engineering Chemistry Paper Solved 2010
    - Advantages of eBooks Over Traditional Books
2. Identifying Engineering Chemistry Paper Solved 2010
    - Exploring Different Genres
    - Considering Fiction vs. Non-Fiction
    - Determining Your Reading Goals
3. Choosing the Right eBook Platform
    - Popular eBook Platforms
    - Features to Look for in an Engineering Chemistry Paper Solved 2010
    - User-Friendly Interface
4. Exploring eBook Recommendations from Engineering Chemistry Paper Solved 2010
    - Personalized Recommendations
    - Engineering Chemistry Paper Solved 2010 User Reviews and Ratings

13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development
    - Exploring Educational eBooks
14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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