

Calculate the Lattice Enthalpy of CaCl_2 crystal

(a) Using Born Hyber Cycle

(b) Using Kapunstisnskii Equation

$$\Delta H_L = -1.21 \frac{nZ_A Z_B}{d} \left(1 - \frac{d^*}{d} \right) \quad \text{MJ} \cdot \text{\AA} \cdot \text{mol}^{-1}$$

Ionic radii: Ca^{+2} : 1.0 \AA ; Cl^- : 1.67 \AA

Enthalpy of formation for $\text{CaCl}_2 = -795 \text{ kJ}$

Sublimation of Ca : 325 kJ

Dissociation enthalpy of Cl_2 : 244 kJ

Electron Affinity for Cl: 355 kJ

Total Ionization enthalpy for $\text{Ca(g)} \rightarrow \text{Ca}^{+2}(\text{g})$: 1735 kJ

$d^* = 0.345 \text{ \AA}$

Enthalpy Of Solution Cacl

Allan Myerson



Enthalpy Of Solution Cacl:

A-level Chemistry E. N. Ramsden, 2000 Each topic is treated from the beginning without assuming prior knowledge Each chapter starts with an opening section covering an application These help students to understand the relevance of the topic they are motivational and they make the text more accessible to the majority of students Concept Maps have been added which together with Summaries throughout aid understanding of main ideas and connections between topics Margin points highlight key points making the text more accessible for learning and revision Checkpoints in each chapter test students understanding and support their private study

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Handbook of Industrial Crystallization Allan Myerson, 2002-01-08 Crystallization is an important separation and purification process used in industries ranging from bulk commodity chemicals to specialty chemicals and pharmaceuticals In recent years a number of environmental applications have also come to rely on crystallization in waste treatment and recycling processes The authors provide an introduction to the field of newcomers and a reference to those involved in the various aspects of industrial crystallization It is a complete volume covering all aspects of industrial crystallization including material related to both fundamentals and applications This new edition presents detailed material on crystallization of biomolecules precipitation impurity crystal interactions solubility and design Provides an ideal introduction for industrial crystallization newcomers Serves as a worthwhile reference to anyone involved in the field Covers all aspects of industrial crystallization in a single complete volume

The NBS Tables of Chemical Thermodynamic Properties Donald D. Wagman, 1982

Physical Chemistry and Industrial Application of Gellan Gum K. Nishinari, 2003-07-01 Gellan gum a microbial polysaccharide consisting of tetra saccharide unit glucose glucuronic acid glucose and rhamnose forms a transparent gel which is heat resistant in the presence of divalent cations Since 1989 the collaborative research group was organised in the Research Group of Polymer Gels affiliated to the Society of Polymer

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Modern Experiments for Introductory College Chemistry Howard A. Neidig,William Franklin Kieffer,1967

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