

MCAT PRACTICE TEST 92 QUESTIONS AND ANSWERS 2024

Consider a biochemical reaction A to B, which is catalyzed by A-B dehydrogenase. Which of the following statements is true?

- A. The reaction will proceed until the enzyme concentration decreases.
 - B. The reaction will be more favorable at 0 degrees C.
 - C. A component of the enzyme transferred from A to B.
 - D. The free energy change (ΔG) of the catalyzed reaction is the same as for the uncatalyzed reaction.
- D.
- In an enzyme-catalyzed reaction, the rate of a reaction is increased by a decrease in the activation energy. Furthermore, enzymes are not changed or consumed during the course of the reaction. Also, the overall free energy change of the reaction, ΔG , remains unchanged in the presence of an enzyme.

Which of the following statements about enzyme kinetics is false?

- A. An increase in the substrate concentration (at constant enzyme concentration) leads to proportional increases in the rate of the reaction.
- B. Most enzymes operating in the human body work best at a temperature of 37 degree C.
- C. An enzyme-substrate complex can either form a product or dissociate back into the enzyme and substrate.
- D. Maximal activity of many human enzymes occurs around pH 7.2.

A.

-Most enzymes in the human body operate a maximal activity around a temperature of 37 degrees C and a pH of 7.2, which is the pH of most body fluids. In addition, as characterized by the Michaelis-Menten model, enzymes form an enzyme-substrate complex, which can either dissociate back into the enzyme and substrate or proceed to form a product. We can eliminate B, C, and D. An increase in the substrate concentration, while maintaining a constant enzyme concentration, leads to a proportional increase in the rate of the reaction only initially. However, once most of the active sites are occupied, the reaction rate levels off, regardless of further increases in substrate concentration. At high concentrations of substrate, the reaction rate approaches its maximal velocity and is no longer changed by further increases in substrate concentration. Therefore statement A is not entirely true.

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as no₃ io₃ re io₃ h₂aso₄

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