

## Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

| Property                         | Acids               | Bases    |
|----------------------------------|---------------------|----------|
| Conductivity in Aqueous Solution | Very                | Very     |
| Taste                            | Sour                | Bitter   |
| Feel                             | —                   | Slippery |
| Reaction with Metal              | produces $H_2$ gas  | —        |
| Reaction with Carbonate          | produces $CO_2$ gas | —        |
| Colour on Litmus Paper           | Red                 | Blue     |
| Colour with Universal Indicator  | Red                 | Purple   |
| Colour with phenolphthalein      | Colourless          | pink     |

2. Indicate if each compound is an acid or base and name it.

|            | Acid/Base? | Name                        |
|------------|------------|-----------------------------|
| $HNO_3$    | <u>A</u>   | <u>nitric acid</u>          |
| $Mg(OH)_2$ | <u>B</u>   | <u>magnesium hydroxide</u>  |
| $H_3PO_4$  | <u>A</u>   | <u>phosphoric acid</u>      |
| $H_2CO_3$  | <u>A</u>   | <u>carbonic acid</u>        |
| HF         | <u>A</u>   | <u>hydrofluoric acid</u>    |
| $Fe(OH)_3$ | <u>B</u>   | <u>iron (III) hydroxide</u> |

3. Write the formula for each of the following acids/bases:

|                    |                              |                   |                              |
|--------------------|------------------------------|-------------------|------------------------------|
| hydrochloric acid  | <u>HCl</u>                   | hydrochloric acid | <u>HCl</u>                   |
| aluminum hydroxide | <u><math>Al(OH)_3</math></u> | sodium hydroxide  | <u>NaOH</u>                  |
| lithium hydroxide  | <u>LiOH</u>                  | calcium hydroxide | <u><math>Ca(OH)_2</math></u> |
| sulfuric acid      | <u><math>H_2SO_4</math></u>  | sulfurous acid    | <u><math>H_2SO_3</math></u>  |

# Acid Base Theories Section Review Answer

**Scott C. Dulebohn**



**Acid Base Theories Section Review Answer:**

## Unveiling the Energy of Verbal Art: An Mental Sojourn through **Acid Base Theories Section Review Answer**

In some sort of inundated with displays and the cacophony of fast connection, the profound energy and psychological resonance of verbal artistry often disappear in to obscurity, eclipsed by the constant assault of sound and distractions. However, nestled within the lyrical pages of **Acid Base Theories Section Review Answer**, a captivating function of fictional splendor that pulses with organic feelings, lies an remarkable journey waiting to be embarked upon. Penned with a virtuoso wordsmith, this magical opus courses readers on a mental odyssey, softly exposing the latent potential and profound influence stuck within the intricate web of language. Within the heart-wrenching expanse with this evocative evaluation, we will embark upon an introspective exploration of the book is central themes, dissect its captivating publishing design, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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