

# STUDY GUIDE

## SECTION 1.1 - 1.4

- **matter**: everything that takes up space + has a mass

### Properties of matter



### ① Physical changes:

- Phase
- Size
- Texture

→ Reversible

### ② Chemical changes:

- Color change
- New substance is formed

→ Irreversible

### Pure substance vs. Mixture

#### MOLECULE (pure substances)

##### Compounds

2 or more  $\neq$  atoms  
(eg:  $H_2O$ ;  $Na_2O$ )

##### Elements

1 atom  
(eg:  $H_2$ ,  $O_2$ )

- **Pure substance**: constant composition + definite properties

# Chemistry Midterm Study Guide

**Clemens Wendtner**



## **Chemistry Midterm Study Guide:**

## Whispering the Techniques of Language: An Mental Quest through **Chemistry Midterm Study Guide**

In a digitally-driven world wherever screens reign supreme and quick conversation drowns out the subtleties of language, the profound secrets and emotional subtleties concealed within words usually go unheard. However, situated within the pages of **Chemistry Midterm Study Guide** a captivating fictional prize pulsating with organic feelings, lies an extraordinary journey waiting to be undertaken. Written by an experienced wordsmith, that enchanting opus attracts visitors on an introspective trip, lightly unraveling the veiled truths and profound impact resonating within the very material of each and every word. Within the mental depths of this emotional evaluation, we will embark upon a heartfelt exploration of the book is key subjects, dissect their charming writing fashion, and fail to the effective resonance it evokes strong within the recesses of readers hearts.

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