

Conceptual Physics
Chapter 17: The Atomic Nature of Matter
Lab 17.2: Building an Atom (PHET simulation)



PART I: ATOM SCREEN

- Go to the website: phet.colorado.edu. Click on HTML5 simulations on top right of screen and choose the Build an Atom simulation (<http://phet.colorado.edu/en/simulation/build-an-atom>)
- Explore the Build an Atom simulation with your group. As you explore, talk about what you find. List two things your group observed in the simulation.
 -
 -
- Click on the + sign for each of the boxes (element name, net charge and mass number) to view changes as you change the number of particles in the atom.
- What particle(s) are found in the center of the atom? _____
- Play until you discover which **p**article(s) determine(s) the name of the **e**lement you build. _____
- What is the name of the following atoms?
 - An atom with 3 protons and 4 neutrons: _____
 - An atom with 2 protons and 4 neutrons: _____
 - An atom with 4 protons and 4 neutrons: _____
- Play with the simulation to discover which particles affect the **charge** of an atom or ion. _____
- Fill in the blanks below to show your results:
 - Neutral atoms have _____ the same number of _____ protons and electrons.
 - Positive ions have _____ protons than electrons.
 - Negative ions have _____ protons than electrons.
- Develop a relationship (in the form of a single sentence or equation) that can predict the charge based on the number and types of particle. _____
- Play with the simulation to discover what affects the **mass** number of your atom or ion. _____
 - What is a rule for determining the mass number of an atom or ion?
- Practice applying your understanding by playing 1st and 2nd levels on the game screen.

Conceptual Physics Chapter 17 Answers

Kathleen Armour



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Table of Contents Conceptual Physics Chapter 17 Answers

1. Understanding the eBook Conceptual Physics Chapter 17 Answers
 - The Rise of Digital Reading Conceptual Physics Chapter 17 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Conceptual Physics Chapter 17 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Conceptual Physics Chapter 17 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Conceptual Physics Chapter 17 Answers
 - Personalized Recommendations
 - Conceptual Physics Chapter 17 Answers User Reviews and Ratings
 - Conceptual Physics Chapter 17 Answers and Bestseller Lists
5. Accessing Conceptual Physics Chapter 17 Answers Free and Paid eBooks
 - Conceptual Physics Chapter 17 Answers Public Domain eBooks

- Conceptual Physics Chapter 17 Answers eBook Subscription Services
- Conceptual Physics Chapter 17 Answers Budget-Friendly Options
- 6. Navigating Conceptual Physics Chapter 17 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Conceptual Physics Chapter 17 Answers Compatibility with Devices
 - Conceptual Physics Chapter 17 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Conceptual Physics Chapter 17 Answers
 - Highlighting and Note-Taking Conceptual Physics Chapter 17 Answers
 - Interactive Elements Conceptual Physics Chapter 17 Answers
- 8. Staying Engaged with Conceptual Physics Chapter 17 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Conceptual Physics Chapter 17 Answers
- 9. Balancing eBooks and Physical Books Conceptual Physics Chapter 17 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Conceptual Physics Chapter 17 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Conceptual Physics Chapter 17 Answers
 - Setting Reading Goals Conceptual Physics Chapter 17 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Conceptual Physics Chapter 17 Answers
 - Fact-Checking eBook Content of Conceptual Physics Chapter 17 Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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