

**LESSON** **Practice B**  
**6-6 Properties of Kites and Trapezoids**

In kite  $ABCD$ ,  $m\angle BAC = 35^\circ$  and  $m\angle BCD = 44^\circ$ .  
 For Exercises 1–3, find each measure.



1.  $m\angle ABD$  \_\_\_\_\_  
 2.  $m\angle DCA$  \_\_\_\_\_  
 3.  $m\angle ABC$  \_\_\_\_\_

4. Find the area of  $\triangle EFG$ . \_\_\_\_\_



5. Find  $m\angle Z$ . \_\_\_\_\_



6.  $KM = 7.5$ , and  $NM = 2.6$ . Find  $LN$ . \_\_\_\_\_



7. Find the value of  $n$  so that  $PQRS$  is isosceles.  
 \_\_\_\_\_



8. Find the value of  $x$  so that  $EFGH$  is isosceles.  
 \_\_\_\_\_



9.  $BD = 7a - 0.5$ , and  $AC = 5a + 2.3$ . Find the value of  $a$  so that  $ABCD$  is isosceles.  
 \_\_\_\_\_



10.  $QS = 8z^2$ , and  $RT = 6z^2 + 38$ . Find the value of  $z$  so that  $QRST$  is isosceles.  
 \_\_\_\_\_



Use the figure for Exercises 11 and 12. The figure shows a ziggurat. A ziggurat is a stepped, flat-topped pyramid that was used as a temple by ancient peoples of Mesopotamia. The dashed lines show that a ziggurat has sides roughly in the shape of a trapezoid.



11. Each "step" in the ziggurat has equal height. Give the vocabulary term for  $\overline{MN}$ .  
 \_\_\_\_\_

12. The bottom of the ziggurat is 27.3 meters long, and the top of the ziggurat is 11.6 meters long. Find  $MN$ .  
 \_\_\_\_\_

# Holt Geometry Chapter Test Answer Key

**N Colangelo**



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web a determine the reaction order for hci using calculations described in the background section show your work note that your answer will probably not be an even whole

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web this set of chemical reaction engineering multiple choice questions answers mcqs focuses on reaction order and elementary rate law 1 molecularity refers to an

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web the rate law for a particular reaction is rate  $k x^2$  in an experiment the initial rate of the reaction is determined to be 0.080 mol l s when the initial concentration of x is

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web a rate law shows how the rate of a chemical reaction depends on reactant concentration for a reaction such as  $a \rightarrow b$  products the rate law generally has the form rate  $k a^n$

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web the reaction orders in a rate law describe the mathematical dependence of the rate on reactant concentrations referring to the generic rate law above the reaction is  $m$

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web each reactant in the rate limiting step of a reaction is assigned an order typically zeroth 1st or 2nd the order is the sum of all individual orders the rate of a reaction is

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