

1. What is the term for metabolic pathways that release stored energy by breaking down complex molecules?

 - A. Anabolic pathways
 - B. Catabolic pathways
 - C. Fermentation process
 - D. Thermodynamic pathways
 - E. Bio energetic pathways
2. The molecule that functions as the reducing agent (electron donor) in a redox or oxidation-reduction reaction

 - A. Gains electron and gains energy
 - B. Loses electrons and loses energy
 - C. Gains electrons and loses energy
 - D. Loses electrons and gains energy
 - E. Neither gain nor loses electrons, but gains or loses energy
3. When electrons move closer to a more electronegative atom, what happens?

 - A. Energy is released
 - B. Energy is consumed
 - C. The more electronegative atom is reduced
 - D. The more electronegative atom is oxidized
 - E. A and C are correct
4. Why does the oxidation of organic compounds by molecular oxygen to produce CO₂ and water release free energy?

 - A. The covalent bond in organic molecules are higher energy bonds than those in water and carbon dioxide
 - B. Electrons are being moved from atoms that have a lower affinity for electrons (such as C) to atoms with a higher affinity for electrons (such as O)
 - C. The oxidation of organic compounds can be used to make ATP
 - D. the electrons have a higher potential energy when associated with water and CO₂ than they do in organic compounds
 - E. The covalent bonds in O₂ is unstable and easily broken by electron from organic molecules
5. Which of the following statements describes the results of this reaction?
 $C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O + \text{energy}$

 - A. C₆H₁₂O₆ is oxidized and O₂ is reduced
 - B. O₂ is oxidized and H₂O is reduced
 - C. CO₂ is reduced and O₂ is oxidized
 - D. C₆H₁₂O₆ is reduced and CO₂ is oxidized
 - E. O₂ is reduced and CO₂ is oxidized
6. When a glucose molecule loses a hydrogen atom as a result of oxidation-reduction reaction, the molecule becomes

 - A. Dehydrogenated
 - B. Hydrogenated
 - C. Oxidized
 - D. Reduced
 - E. An oxidizing agent
7. When a molecule of NAD⁺ (nicotinamide adenine dinucleotide) gains a hydrogen atom (not a hydrogen ion) the molecule becomes

 - A. Hydrogenated
 - B. Oxidized
 - C. Reduced
 - D. Redoxed
 - E. A reducing agent
8. Which of the following statements describes NAD⁺?

 - A. NAD⁺ is reduced to NADH during both glycolysis and the citric acid cycle
 - B. NAD⁺ has more chemical energy than NADH
 - C. NAD⁺ is reduced by the action of hydrogenases
 - D. NAD⁺ can donate electrons for use in oxidative phosphorylation
 - E. In the absence of NAD⁺, glycolysis can still function
9. Where does glycolysis take place?

 - A. Mitochondrial matrix
 - B. Mitochondrial outer membrane
 - C. Mitochondrial inner membrane
 - D. Mitochondrial intermembrane space
 - E. Cytosol
10. The ATP made during glycolysis is generated by

 - A. Substrate-level phosphorylation
 - B. Electron transport
 - C. Photo phosphorylation
 - D. Chemiosmosis
 - E. Oxidation of NAD⁺ and NADH
11. The oxygen consumed during cellular respiration is involved directly in which process or event?

 - A. Glycolysis
 - B. Accepting electrons at the end of the electron transport chain
 - C. The citric acid cycle
 - D. The oxidation of pyruvate at acetyl CoA
 - E. The phosphorylation of ADP to form ATP
12. Which process in eukaryote cells will proceed normally whether oxygen (O₂) is present or absent?

 - A. Electron transport
 - B. Glycolysis
 - C. The citric acid cycle
 - D. Oxidation phosphorylation
 - E. Chemiosmosis

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Mark Anestis, Kelcey Burris



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