

## AP Chemistry Chemical Kinetics Worksheet Answers

1.	Half lives	Reactant Fraction Remaining
	1	$\frac{1}{2}$
	2	$\frac{1}{4}$
	3	$\frac{1}{8}$
	4	$\frac{1}{16}$
	5	$\frac{1}{32}$

2. 2<sup>nd</sup> Order; 1<sup>st</sup> Order

3. a. True    b. True

c. False; As [Reactant] decreases the rate will decrease    d. True

4. a.  $-1/2 \Delta[\text{O}_3]/\Delta t = 1/3 \Delta[\text{O}_2]/\Delta t$   
b.  $-1/2 \Delta[\text{HOF}]/\Delta t = 1/2 \Delta[\text{HF}]/\Delta t = \Delta[\text{O}_2]/\Delta t$   
c.  $-\Delta[\text{N}_2]/\Delta t = -1/3 \Delta[\text{H}_2]/\Delta t = 1/2 \Delta[\text{NH}_3]/\Delta t$

5. a. rate =  $k[\text{NO}_2]^2[\text{CO}]^0$  or  $k[\text{NO}_2]^2$   
b. rate = 1/4 the original value

6. The initial rate  $R = k[\text{B}]_0^2$

7. a.  $6.0 \times 10^{-3} / 1.2 \times 10^{-2} = (0.10/0.20)^X$   
 $X = 1$  so it is 1<sup>st</sup> order with respect to A  
 $1.1 \times 10^{-1} / 1.2 \times 10^{-2} = (0.15/0.05)^X$   
 $X = 2$  so it is 2<sup>nd</sup> order with respect to B  
**Rate Law =  $k[\text{A}][\text{B}]^2$**

b. Overall it is 3<sup>rd</sup> order

c. Probably not a single step since a 3-body collision would have to occur.

8. a. half life =  $\ln(1/2) / -5.0 \times 10^{-4} \text{ s}^{-1} = 1.4 \times 10^3 \text{ s}$   
 $-\ln(0.5) / 5.0 \times 10^{-4} \text{ s}^{-1} = 1.4 \times 10^3 \text{ s}$   
b. Assign an original value of 1.00M  
 $\ln(0.1/1.0) = -5.0 \times 10^{-4} \text{ s}^{-1} t$   
 $t = -\ln(0.1) / 5.0 \times 10^{-4} \text{ s}^{-1} = 4.6 \times 10^3 \text{ s}$

9. After 30.min (one half-life)

$$P_{\text{HOF}} = \frac{1}{2} (100. \text{mmHg}) = 50. \text{mmHg}$$

$$P_{\text{HF}} = 50. \text{mmHg}$$

$$P_{\text{O}_2} = 25. \text{mmHg}$$

Calculating the rate constant  $k$ ,

$$\ln([\text{HOF}]_t / [\text{HOF}]_0) = -kt \quad \ln(0.5) = -k(30. \text{min})$$

# Ap Chemistry Chemical Kinetics Worksheet Answers

**Eli Usheunepa Yunana**



**Ap Chemistry Chemical Kinetics Worksheet Answers:**

## **Ap Chemistry Chemical Kinetics Worksheet Answers** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Ap Chemistry Chemical Kinetics Worksheet Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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