

# CHIMICA GENERALE

Ogni atomo/elemento è caratterizzato da una simbolo

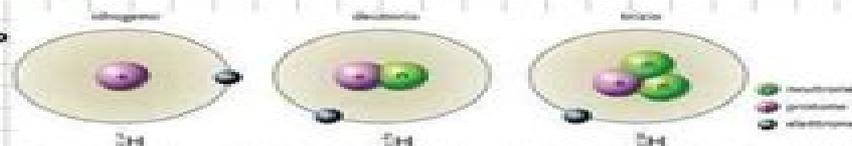


- il numero di massa è dato dalla somma di protoni + neutroni (nucleoni)
- il numero atomico corrisponde al numero dei protoni (uguale al numero di e<sup>-</sup>)

un nucleo con particolari valori di Z e A → **NUCLIDE**

**GLI ISOTOP:**

Atomi di uno stesso elemento che possono differire per il numero di neutroni. I diversi isotopi di un elemento hanno uguali proprietà chimiche, ma diverse proprietà fisiche.



→ **CATIONI:** quando ad un atomo vengono sottratti uno o più elettroni, si formano ioni carichi positivamente.

→ **ANIONI:** quando l'atomo assume uno o più elettroni si formano ioni carichi negativamente.

- contiene 7 periodi (orizzontali)
- 18 gruppi (verticali) → nei gruppi gli elementi hanno caratteristiche chimiche simili

**TAVOLA DEGLI ELEMENTI:**

essi sono organizzati per massa atomica crescente.

gruppi	1A	2A	3A	4A	5A	6A	7A	8A	9A	10A	11A	12A	13A	14A	15A	16A	17A	18A
1	H	He																
2	Li	Be																
3	Na	Mg	Al	Si	P	S	Cl	Ar										
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
6	Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
7	Fr	Ra																

■ orbitale s  
■ orbitale p  
■ orbitale d  
■ orbitale f

- **GRUPPO 1:** metalli alcalini (Li, Na, K, Rb, Cs)
- **GRUPPO 2:** metalli alcalino-terrosi (Be, Mg, Ca, Sr, Ba)
- essi reagiscono con l'acqua producendo idrogeno e calore es il potassio reagisce tanto da fiammare l'idrogeno.
- **GRUPPO 18:** gas nobili, essi si combinano con pochissimi altri elementi. Sono tutti gas incolori e inodori. Sono poco reattivi.
- **GRUPPO 17:** alogeni. Si presentano con molecole biatomiche.

- **I METALLI:** sono solidi, dotati di lucentezza, malleabilità e duttilità; essi sono buoni conduttori di elettricità e calore.

- **I NON METALLI:** sono gas o solidi che non hanno caratteristiche metalliche.

- **SEMIMETALLI O METALLOIDI:** hanno caratteristiche intermedie tra i metalli e i non metalli.

# Appunti Di Chimica Generale Unife

**Riccardo Ciusa**



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In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Appunti Di Chimica Generale Unife** a literary masterpiece penned with a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those that partake in its reading experience.

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