



Separate Chemistry

What is a substance named?

What are examples of these substances named?

What are the following named as one of metals + oxides or non-metals + oxides?

What are the following named as?

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Why are hydrogen and carbon compounds considered as the reducing agent?

What are the two methods used when hydrogen and carbon reducing?

Why is a splint held over the gas test?

Complete the acid reactions

acid + carbonate → carbon dioxide +

acid + hydroxide → hydrogen oxide +

acid + metal → salt +

acid + hydrogen sulfide →

Describe what a metal reacting with an acid can tell you about the reactivity of the metal.



What are the products?

- metal salt
- hydrogen gas
- metal oxide
- metal sulfide
- metal hydroxide
- metal carbonate

What are the products?



How much heat will water produce?

How much heat will acid produce?

Is a reaction of gas or not. (solid heat)

- strong oxidant
- strong acid
- water
- hydrogen gas
- metal salt
- gas

Describe how to make a soluble salt from an insoluble base

1. Weigh out a _____
2. Weigh out a _____ salt
3. Wash the _____
4. Add the insoluble base to the acid until there is no further _____
5. _____ the mixture
6. Heat the solution to a _____ the water
7. _____ if you want a pure salt

Complete the acid reactions



What are the products of a carbonate reaction?



Complete the following

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____

_____ is the acid formed and _____ is the salt of the acid

Investigation: Separating a mixture of ions

1. Weigh out a _____

2. Weigh out a _____ salt

3. Wash the _____

4. Add the insoluble base to the acid until there is no further _____

5. _____ the mixture

6. Heat the solution to a _____ the water

7. _____ if you want a pure salt

8. _____ the mixture

9. _____ the mixture

10. _____ the mixture

11. _____ the mixture

12. _____ the mixture

13. _____ the mixture

14. _____ the mixture

15. _____ the mixture

16. _____ the mixture

17. _____ the mixture

18. _____ the mixture

19. _____ the mixture

20. _____ the mixture

Investigation: Separating a mixture of ions

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12. _____ the mixture

13. _____ the mixture

14. _____ the mixture

15. _____ the mixture

16. _____ the mixture

17. _____ the mixture

18. _____ the mixture

19. _____ the mixture

20. _____ the mixture



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Samantha Andrews



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while lonesome suzie is a curtis mayfield ballad soaked in tranquilized

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