

0 9 . **4** In an experiment starting with 5.05 g of phenylamine, 4.82 g of purified product were obtained in step 1.

Calculate the percentage yield in this reaction.
Give your answer to the appropriate number of significant figures.

[3 marks]

Percentage yield = _____ %

0 9 . **5** A reagent for step 2 is a mixture of concentrated nitric acid and concentrated sulfuric acid, which react together to form a reactive intermediate.

Write an equation for the reaction of this intermediate in step 2.

[1 mark]

0 9 . **6** Name a mechanism for the reaction in step 2.

[1 mark]

0 9 . **7** Suggest the type of reaction occurring in step 3.

[1 mark]

0 9 . **8** Identify the reagents used in step 4.

[1 mark]

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Janet Helm, Arthur Robinson



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