

# AQA A-level Biology

## Condensed notes

### MONOMERS & POLYMERS

- Monomers - the smaller units from which large molecules are made
- Polymers - molecules made from many monomers of identical or similar structures
- Condensation reaction - a reaction where bonds are formed and water is released
- Hydrolysis reaction - a reaction where bonds are broken and water is released

Polymers	Monomers	Bond
polysaccharide	monosaccharide	glycosidic bond
protein	amino acid	peptide bond
Nucleic acid e.g. DNA / RNA	nucleotide	phosphodiester bond

\*Small and very similar monomers that can be made of repeating units

### INORGANIC IONS

- Simplest ion - an ion that doesn't contain other ions
- Can be an anion (negative) or cation (positive) charge
- Often not with negative charge
- Often not with positive charge

Ion	Role
<b>Hydrogen ions (H<sup>+</sup>)</b>	<ul style="list-style-type: none"> <li>involved in many biological processes</li> <li>control the pH of a solution</li> <li>act as a catalyst in many biological reactions</li> <li>act as a cofactor in many enzymes</li> <li>act as a signal in many biological processes</li> </ul>
<b>Calcium ions (Ca<sup>2+</sup>)</b>	<ul style="list-style-type: none"> <li>essential for bone structure</li> <li>involved in muscle contraction</li> <li>act as a cofactor in many enzymes</li> </ul>
<b>Magnesium ions (Mg<sup>2+</sup>)</b>	<ul style="list-style-type: none"> <li>essential for chlorophyll synthesis</li> <li>act as a cofactor in many enzymes</li> <li>involved in muscle contraction</li> </ul>
<b>Iron ions (Fe<sup>2+</sup>)</b>	<ul style="list-style-type: none"> <li>essential for haem synthesis</li> <li>act as a cofactor in many enzymes</li> <li>involved in electron transport</li> </ul>
<b>Zinc ions (Zn<sup>2+</sup>)</b>	<ul style="list-style-type: none"> <li>essential for many enzymes</li> <li>act as a cofactor in many enzymes</li> <li>involved in muscle contraction</li> </ul>

### NUCLEIC ACIDS

**NUCLEIC ACIDS**  
 Polymers made from **monomers** called **NUCLEOTIDES**.  
 Polymers called **NUCLEIC ACIDS** contain **phosphate** and **sugar** and **nitrogenous bases**.  
 DNA and RNA are both **polynucleotides**.  
 They are made using the same rule for the formation of polymers.



- Monomers are polynucleotides - a polynucleotide is a long chain of nucleotides
- The order of nitrogenous bases in the sugar phosphate backbone
- Hydrolysis - an enzyme that can break down a polymer into its monomers

**NUCLEOTIDES**  
 Each nucleotide consists of a phosphate group, a pentose sugar and a nitrogenous base.  
 The phosphate group is attached to the 5' carbon of the pentose sugar.  
 The nitrogenous base is attached to the 1' carbon of the pentose sugar.  
 The pentose sugar is attached to the 3' carbon of the pentose sugar.



### CARBOHYDRATES

**CARBOHYDRATES**  
 All carbohydrates contain the elements **C, H and O**.  
 Monosaccharides are the **simplest** form of carbohydrate and have the general formula **C<sub>n</sub>H<sub>2n</sub>O<sub>n</sub>**.  
 Disaccharides are made from two monosaccharides.  
 Polysaccharides are made from many monosaccharides.



**MONOSACCHARIDES**  
 The simplest form of carbohydrate. They are made from the elements C, H and O.  
 They have the general formula C<sub>n</sub>H<sub>2n</sub>O<sub>n</sub>.  
 They are made from the elements C, H and O.



- Monosaccharides - the simplest form of carbohydrate
- Disaccharides - made from two monosaccharides
- Polysaccharides - made from many monosaccharides
- Hydrolysis - an enzyme that can break down a polymer into its monomers
- Condensation reaction - a reaction where bonds are formed and water is released
- Hydrolysis reaction - a reaction where bonds are broken and water is released

**MONOSACCHARIDES**  
 The simplest form of carbohydrate. They are made from the elements C, H and O.  
 They have the general formula C<sub>n</sub>H<sub>2n</sub>O<sub>n</sub>.  
 They are made from the elements C, H and O.

# Biological Molecules

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**Patrick Vollmar**



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