

Chapter Test A

Chapter: Chemical Equations and Reactions

Do the work carefully, using the letter of the letter of phrases that best completes each statement or best answers each question.

1. The atoms contained in both reactants in a balanced chemical equation following the procedure described in this chapter always give the same structural formula for reactants.
 - a. They become intermixed.
 - b. They are completely used up.
 - c. They are rearranged to form the products, which are different.
 - d. None of the above; all three given things are true simultaneously.
2. In the reaction described by a balanced equation:

$$\text{reactants} \rightarrow \text{products} + \text{products} + \text{products} + \text{products}$$
 - a. all the reactants are products.
 - b. all the reactants are products.
 - c. all the reactants are products.
 - d. None is used up.
3. Which of the following relationships is always true for a balanced equation?
 - a. The products of a reaction are equal in mass to the reactants (as found prior) of the reaction.
 - b. All of the reactants are products.
 - c. The reactants of a balanced equation equal the products of a reaction.
 - d. The reactants of a balanced equation are equal in mass to the products of a reaction.
4. There is a negative enthalpy value associated with a reaction. Which of the following is true?
 - a. The reactants of the reaction are products.
 - b. All the reactants are products.
 - c. The products of the reaction are products.
 - d. All of the above.
5. In a reaction, a chemical equation is balanced.
 - a. equal the reactants.
 - b. equal the products.
 - c. equal the reactants and products.
 - d. equal the reactants.

Chapter 9 Modern Chemistry Test

L Cohen



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