

Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON_METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.) H ₂ hydrogen is diatomic: $\text{H} + \text{H} \longrightarrow \text{H} \cdot \cdot \text{H} \longrightarrow \text{H}-\text{H} \quad \text{Single Bond}$
2.) F ₂ fluorine is diatomic: $\text{F} + \text{F} \longrightarrow \cdot \cdot \text{F} \cdot \cdot \cdot \cdot \text{F} \cdot \cdot \longrightarrow$
3.) O ₂ oxygen is diatomic: $\text{O} + \text{O} \longrightarrow \quad \quad \quad \longrightarrow \text{O}=\text{O} \quad \text{Double Bond}$
4.) N ₂ nitrogen is diatomic. Is this a triple bond? $\text{N} + \text{N}$
5.) BF ₃ you need 3 fluorine atoms here: $\text{B} + \text{F}$
6.) Ammonia NH ₃ hint: how many hydrogen atoms are needed? $\text{N} + \text{H}$
7.) Carbon dioxide CO ₂ $\text{O} + \text{C} + \text{O}$
8.) Methane CH ₄ careful here 4 hydrogen atoms needed $\text{C} + \text{H}$
9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!! $\text{H} + \text{O} + \text{H}$
10.) SO ₂ hint: one pair of electrons from sulfur must be slit up for this one to work. $\text{O} + \text{S} + \text{O}$

Chapter 8 Covalent Bonding Worksheet Answers

G Orfield



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Unveiling the Magic of Words: A Report on "**Chapter 8 Covalent Bonding Worksheet Answers**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Chapter 8 Covalent Bonding Worksheet Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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