



## The Chemistry of Life

## Chapter Test A

### Multiple Choice

Directions: Select the best answer for each of the questions by circling the number of the best possible answer.

1. If an atom has a mass number of 23 and 11 electrons, the atom is most likely to
  - a. be an ion
  - b. be neutral
  - c. be a cation
  - d. be an anion
2. If an atom has a mass number of 23 and 11 protons, the atom is most likely to
  - a. be an ion
  - b. be neutral
  - c. be a cation
  - d. be an anion
3. Which of the following is not a characteristic of water?
  - a. low density of hydrogen-bonded water molecules
  - b. very strong surface tension and surface adhesion
  - c. moderate solubility and high capacity of hydrogen
  - d. moderate solubility and low density of water
4. A molecule that is neutral in the standard
  - a. bonding condition
  - b. bonding process
  - c. bonding electron
  - d. bonding process
5. Which molecule is polar, with the
  - a. oxygen with being slightly positive and the hydrogen with being slightly negative
  - b. oxygen with hydrogen with being slightly positive
  - c. oxygen with hydrogen with being slightly negative
  - d. oxygen with being slightly negative and the hydrogen with being slightly positive
6. If you are with water, which is the most
  - a. positive with hydrogen
  - b. positive with oxygen
  - c. positive with electron
  - d. positive with
7. A substance with a pH of 1 is called
  - a. an acid
  - b. a base
  - c. a salt
  - d. a compound
8. Sodium chloride has a pH of 7. It is
  - a. neutral
  - b. basic
  - c. acidic
  - d. neutral
9. Why is coffee considered compared to other drinks?
  - a. Coffee is a very acidic and has a lot of different substances
  - b. Coffee is a very basic and has a lot of different substances
  - c. Coffee is a very acidic and has a lot of different substances
  - d. Coffee is a very basic and has a lot of different substances

# Chapter Test Chemistry Of Life Answer Key

**Sebastian Brünink**



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