

Chemistry Final 2013 – Semester 1 Review Guide

I. CLASSIFICATION OF MATTER

1. Define the following:

- a. Elements: _____
- b. Compound: _____
- c. Mixture: _____

2. What is the difference between homogeneous and heterogeneous mixtures?

3. Identify the following as an element (e), compound (c), or mixture (m):

Substance	E, C or M
As	
NaCl	
Sand + Water	
S	
Air	

4. Which of the following indicate a chemical change:

- (a) Fizzing
(b) Burning
(c) Color change
(d) All of the above

5. Which of the following are physical changes:

- (a) State of changes (solid, liquid, gas)
(b) Dissolving
(c) Cracking open an egg
(d) All of the above

6. What is the difference between a physical change and a chemical change?

II. LABORATORY SAFETY

1. When should long hair be tied back?

2. When chemicals are spilled what should you do?

Chemistry Midterm Exam Review

Mark Steyvers



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